# Control of remote stereochemistry using phosphine oxides: formal synthesis of any stereoisomer diol ( $R R, R S, S R$ or $S S$ ) bearing 1,5-related stereogenic centres across an $E$ double bond 

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$\gamma$-Alkenyl $\beta$-hydroxy phosphine oxides have been epoxidised stereoselectively to give $\gamma, \delta$-epoxy $\beta$-hydroxy phosphine oxides with high anti stereoselectivity. The $\gamma$-anisyl or $\gamma$-furyl ring of $\gamma$-aryl $\beta$-hydroxy phosphine oxides have been cleaved oxidatively to reveal a carboxylic acid and a ketone respectively. In the latter case, the ketone was reduced highly stereoselectively to give ( $4 R^{*}, 5 S^{*}, 6 R^{*}$ )-8-benzyloxy-6-diphenylphosphinoyloctane-1,4,5-triol as a single diastereoisomer with three controlled stereogenic centres. This method was then applied to the synthesis of three of the diastereoisomers of 8-benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol with four controlled stereogenic centres; the middle two stereogenic centres were removed using an $E$-selective Horner-Wittig elimination to give either diastereoisomer of 8-benzyloxydodec-5-ene-1,4-diol with 1,5-related stereogenic centres across an $E$ alkene.

## Introduction

In the preceding paper, ${ }^{1}$ we outlined a general strategy for the synthesis of allylically functionalised compounds ${ }^{2,3}$ with 1,4 related stereogenic centres across double bonds of fixed configuration. ${ }^{4}$ In this paper, we take this strategy one stage further in the synthesis of alkenyl diols with 1,5 -related stereogenic centres across an $E$ double bond. We have shown that the diphenylphosphinoyl group ${ }^{5}$ can be used to control the 1,3,4related stereogenic centres in $\beta$-hydroxy phosphine oxides $\mathbf{1 .}{ }^{1}$ We now describe methods which allow the $\mathrm{R}^{2}$ group of phosphine oxides 1 to be transformed into a prochiral unit (as in 2) which can be functionalised stereoselectivity (for example by alkene epoxidation, ketone reduction) to give phosphine oxides 3 with four stereogenic centres (Scheme 1). Horner-


Wittig elimination will then be used to "harvest" the middle two stereogenic centres to give alkenes $\mathbf{4}$ with a controlled 1,5 chiral relationship. The aim of the work was to develop methods which allow the general synthesis of any stereoisomer of the $\beta$-hydroxy phosphine oxides 3 and hence the alkenes 4.

[^0]

Fig. 1

## Results and discussion

Stereoselective epoxidation of $\gamma, \delta$-unsaturated $\boldsymbol{\beta}$-hydroxy phosphine oxides
Initially, we looked at epoxidation as a means of introducing a $\gamma$ stereogenic centre into molecules of general structure 2 $\left(\mathrm{X}=\mathrm{CH}_{2}\right)$. We chose to use racemic $\beta$-hydroxy phosphine oxides 5 and $\mathbf{8}$ as models of $2\left(\mathrm{X}=\mathrm{CH}_{2}\right)$. Epoxidation of 5 and $\mathbf{8}$ with MCPBA gave the epoxides $\mathbf{6}$ and 9 with high diastereoselectivity and in excellent yield; in both cases, the epoxide oxygen of the products was on the opposite face to the hydroxy group as shown in Scheme 2. Earlier work suggested that the epoxides were formed by directed epoxidation of the conformation $\mathbf{1 2}$ in which 1,2-allylic strain is minimised (Fig. 1). ${ }^{3}$ Epoxides similar to $\mathbf{6}$ and 9 are extremely sensitive to Payne rearrangement, ${ }^{3,6}$ so the crude reaction mixtures were treated directly with lithium benzenethiolate, ${ }^{7}$ in the presence of a PhSH buffer, to give the diols $\mathbf{7}$ and $\mathbf{1 0}$. We also tried unsuccessfully to open the epoxide 6 with the anions of 5bromouracil and 6-chloropurine. ${ }^{8,9}$ Horner-Wittig elimination of the $\beta$-hydroxy phosphine oxide $\mathbf{1 0}$ gave the $Z$-allylic alcohol 11 in poor yield. $\ddagger$
Many features of this epoxidation route were unsatisfactory. To start with, the syntheses of optically active allylic alcohols 1 ( $\mathrm{R}^{2}=$ alkenyl) were neither flexible nor stereoselective. ${ }^{1}$ In any case, precedent suggested that epoxidation of the syn isomers of 6 and 9 would have been low yielding and unselective. ${ }^{3}$ Furthermore, we had been unable to find conditions which

[^1]Table 1 Acylation and oxidative cleavage of the $\beta$-hydroxy phosphine oxides 16, 20 and 24

| Entry | Starting material | Acylation |  |  | Oxidative cleavage |  |  | Transformation |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Conditions ${ }^{\text {a }}$ | $\beta$-acyloxy phosphine oxide | Yield <br> (\%) | Conditions ${ }^{\text {a }}$ | Product | Yield <br> (\%) | Conditions ${ }^{\text {a }}$ | Product | Yield <br> (\%) |
| 1a | 16 | (a) | 17 | 95 | (b) | 18 | 91 | (c) | 19 | 55 |
| 1b |  |  |  |  |  |  |  | (d) | 19 | 44 |
| 2 | 20 | (a) | 21 | 81 | (b) | 22 | ${ }^{\text {b }}$ | (e) | 23 | 18 |
| 3 | 25 | (a) | 25 | >98 | (b) | 26 | >98 | - | - | - |
| 4 | 16 | (f) | 27 | 72 | (b) | 28 | b | (c) | 29 | 59 |
| 5 | 20 | (f) | 30 | 75 | (b) | 31 | ${ }^{\text {b }}$ | (c) | 32 | 55 |
| 6 | - | - | 36 | - | (b) | 37 | $b$ | (c) | 38 | 61 |

${ }^{a}$ Reagents and conditions: (a) PhCOCl , cat. DMAP, $\mathrm{Et}_{3} \mathrm{~N}$; (b) cat. $\mathrm{RuCl}_{3}, \mathrm{NaIO}_{4}, \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeCN}^{2}-\mathrm{H}_{2} \mathrm{O}$; (c) SOCl , MeOH ; (d) $\mathrm{CH}_{2} \mathrm{~N}_{2}$; (e) CDI , MeNHOMe $\cdot \mathrm{HCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; (f) $\mathrm{Ac}_{2} \mathrm{O}$, pyridine. ${ }^{b}$ Crude product was the acid by $400 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectroscopy.


Scheme 2 Reagents and conditions: (a) $M$-CPBA, $\mathrm{Na}_{2} \mathrm{HPO}_{4}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, $0^{\circ} \mathrm{C}$; (b) PhSLi, PhSH, THF; (c) $\mathrm{NaH}, \mathrm{DMF}$.
promoted clean Horner-Wittig elimination of diols like $\mathbf{1 0}$. These problems led us to abandon this strategy.

## Oxidative cleavage of the anisyl ring of $\boldsymbol{\beta}$-anisyl $\boldsymbol{\beta}$-acyloxy phosphine oxides

An alternative strategy for the synthesis of molecules similar to $2\left(\mathrm{X}=\mathrm{O} ; \mathrm{R}^{3}=\mathrm{OH}\right)$ involved the oxidative cleavage of the aromatic ring of phosphine oxides similar to $\mathbf{1}\left(\mathrm{R}^{2}=p-\mathrm{MeOC}_{6} \mathrm{H}_{4}\right)$. To start with, we chose to work with models of the phosphine oxides $1\left(\mathrm{R}^{2}=p\right.$ - $\left.\mathrm{MeOC}_{6} \mathrm{H}_{4}\right)$ such as 13 . The $\beta$-hydroxy phosphine oxide 13 was protected as the dibenzyl ether 14; oxidation of the dibenzyl ether $\mathbf{1 4}$ under Sharpless's conditions ${ }^{10}$ gave only diester 15 in low yield (Scheme 3). Evidently, the benzyl ethers of $\mathbf{1 4}$ are more susceptible to oxidation under these conditions than the anisyl ring.

We protected the $\beta$-hydroxy phosphine oxides 16, 20 and 24 as benzoates $\mathbf{1 7}, 21$ and $\mathbf{2 5}$ and acetates (e.g. $\mathbf{2 7}$ or $\mathbf{3 0}$ ) because these functional groups were known to be stable to oxidation under Sharpless's conditions (Scheme 4; Table 1). ${ }^{10}$ In a similar


Scheme 3
vein, the $\beta$-hydroxy phosphine oxide $\mathbf{1 3}$ was converted into the phthalimide 36 using the chemistry described in Scheme 5; the benzyl ether of $\mathbf{3 4}$ was easily removed by hydrogenolysis and the alcohol of 35 was substituted by phthalimide, using a Mitsunobu reaction. ${ }^{11}$

The benzoates 17, 21 and 25 and the acetates 27, 30 and 36 were oxidised to the corresponding acids using sodium periodate and catalytic ruthenium trichloride in a $1: 1: 1$ mixture of acetonitrile, water and carbon tetrachloride (Schemes 4 and 5; Table 1). Practically, we found that about 15 equivalents of sodium periodate were necessary for the reaction to reach completion; it was also necessary to use considerably more solvent ${ }^{12}$ than that recommended by Sharpless ${ }^{10}$ to allow efficient stirring of the reaction mixture. The crude reaction mixtures were analysed by ${ }^{1} \mathrm{H}$ NMR and were almost exclusively the required carboxylic acids. The efficiency and chemoselectivity of these oxidation reactions is remarkable; in each case, the anisyl ring is selectively oxidised in the presence of three phenyl rings, each of which is protected by a neighbouring carbonyl or phosphinoyl group.

Ideally, we wanted to develop simple and efficient ways to convert the crude acids into useful prochiral units. For example, the acid 18 was converted into the methyl ester 19 by treatment with acidic methanol solution (entry 1) or diazomethane (entry 2, Table 1); the methyl ester 19 was, however, resistant to attack by a variety of organometallic reagents and could be converted in only $52 \%$ yield into the diol 33 using lithium aluminium hydride. We have previously synthesised similar diphenylphosphinoyl diols using the Sharpless asymmetric dihydroxylation reaction. ${ }^{13}$ The acid 22 was converted in low yield into the Weinreb amide 23 using carbonyl diimidazole and $\mathrm{N}, \mathrm{O}$ dimethylhydroxylamine hydrochloride. ${ }^{14}$

Most of the problems associated with protecting $\beta$-hydroxy phosphine oxides as benzoates were easily solved by using an acetate protecting group instead (entries 4-6, Table 1). The


13
34

cat. $\mathrm{RuCl}_{3}, \mathrm{NaIO}_{4}$
$\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeCN}-\mathrm{H}_{2} \mathrm{O}$

Scheme 5


$$
\text { Scheme } 6
$$

we wanted; the furan ring of $\mathbf{4 6}$ had been transformed into a prochiral unit (a ketone) suitable for further functionalisation. Stereoselective functionalisation of the enone 47 was also surprisingly easy; the open-chain triol 49, with its three controlled stereogenic centres, was obtained as a single diastereomer simply by reducing ${ }^{18} 47$ with sodium borohydride. Presumably, opening of the hemiacetal of 47 and reduction of the resulting aldehyde is followed by 1,4-addition and stereoselective 1,2reduction of the enone. ${ }^{* *}$ Available evidence ${ }^{20}$ suggests that such

[^2]
$51 \rightarrow 49$
Fig. 2
reactions are syn selective, and proceed under Felkin control (Fig. 2). ${ }^{21}$
An obvious development of this work was to study the effect of cerium trichloride on the outcome of the reduction of 47. Luche's reduction conditions ${ }^{22}$ are well known to promote 1,2 reduction of $\alpha, \beta$-unsaturated ketones, but the addition of the Lewis-acidic cerium trichloride can also dramatically alter the stereochemical course of reactions. ${ }^{23}$ Reduction of enone 47 using Luche's conditions did indeed leave the double bond intact, and once more a single diastereomer (48) was obtained (Scheme 8). The stereochemistry of $\mathbf{4 8}$ was proved by catalytic hydrogenation to an 85:15 mixture of triol 49 and starting material. Evidently, the reduction of 47 with sodium borohydride in the presence of cerium(III) chloride proceeded with the same stereochemical sense as reduction with sodium borohydride alone.

Horner-Wittig elimination of 49 using potassium hydroxide in DMSO gave the alkene 50, albeit in poor yield. Remarkably, alkene $\mathbf{5 0}$ was obtained as the $(E)$ geometric isomer, indicating that the Horner-Wittig elimination had not been stereospecific. There are many examples of Horner-Wittig eliminations in which the usual $\operatorname{syn}$ stereospecificity has been lost, ${ }^{24}$ and these examples are usually explained by a particularly easy retro-Horner-Wittig addition which can compete with the olefination process. In this case, however, we propose an alternative explanation which builds on other Horner-Wittig eliminations of $\beta, \gamma$-dihydroxy phosphine oxides. ${ }^{13}$ Base-catalysed elimination of 49, to give the vinyl phosphine oxide 52, followed by readdition of hydroxide to give $\mathbf{5 3}$ would provide a means by


Scheme 7


Scheme 8


Scheme 9
which 49 could be converted into 53 under the reaction conditions (Scheme 9). Horner-Wittig elimination of 53 would then give the observed $(E)$-alkene $\mathbf{5 0}$.

## Synthesis of unsaturated diols with 1,5-related stereogenic centres across an $E$ alkene

The stage was now set for the synthesis of some molecules with remote stereogenic centres across a double bond. Oxidation of the furan rings of $\mathbf{5 4}$ and $\mathbf{5 7}$, followed by reduction with sodium borohydride introduced the fourth stereogenic centre of triols

55 and 58 with high stereoselectivity (Scheme 10). Previously, this sequence of reactions had been studied using the anti $\beta$ hydroxy phosphine oxide 46 (Scheme 8); we were pleased to find that reduction was equally stereoselective with syn phosphine oxides 54 and $57 . \dagger \dagger$ Removal of the middle two stereogenic centres of triols $\mathbf{5 5}$ and $\mathbf{5 8}$ by Horner-Wittig elimination was
$\dagger \dagger$ A similarly remote stereogenic centre has been observed to have a remarkable effect on the rate and stereoselectivity of some diphenylphosphinoyl alkenes [ref. 3(a)].


Scheme 10 Reagents and conditions: (a) MCPBA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; (b) $\mathrm{NaBH}_{4}$, EtOH ; (c) KOH , DMSO, $55^{\circ} \mathrm{C}$.
completely $E$-selective, giving unsaturated diols $\mathbf{5 6}$ and $\mathbf{5 9}$ in rather low yield. ${ }^{25}$

This approach was easily extended to cyclohexyl-substituted diols 62 and $65 .{ }^{25}$ Unfortunately, it was not possible to separate the ${ }^{1,3}$ syn and ${ }^{1,3}$ anti isomers $\mathbf{6 0}$ and $\mathbf{6 3}$, so diols $\mathbf{6 2}$ and $\mathbf{6 5}$ were isolated as mixtures. Nevertheless, the 1,5 -relationship between the stereogenic centres of diols $\mathbf{6 2}$ and $\mathbf{6 5}$ was partly (between $2: 1$ and $3: 1$ ) controlled which is a remarkable feat considering the remoteness (in space) of the stereogenic centres.

The conversion of $\mathbf{6 6}$ into the diol $\mathbf{5 6}$ was also studied and, once again, the Horner-Wittig elimination of an anti $\beta$-hydroxy phosphine oxide (68) gave an $E$ alkene (Scheme 11). Nevertheless, the fact that our Horner-Wittig eliminations ignored the relative stereochemistry of the $\alpha$ and $\beta$ stereogenic centres of $\mathbf{5 5}, 58$ and $\mathbf{6 8}$ meant that both ${ }^{1,5}$ syn and ${ }^{1,5}$ anti diols 56 and 59 could be made from the same starting material 66 . This had been possible because 57 was synthesised from $\mathbf{6 6}$ using an oxidation-reduction sequence. ${ }^{1}$

Despite the inevitably $E$-selective Horner-Wittig elimination of the triols $\mathbf{5 5}, 58$ and $\mathbf{6 8}$, the use of single diastereomers throughout the reaction sequences described in Schemes 10 and 11 is still of fundamental importance. The stereochemistry of the $\beta$-hydroxy phosphine oxide unit of triols $\mathbf{5 5}, \mathbf{5 8}$ and $\mathbf{6 8}$ still controlled the gradual transmission of stereochemical information from the stereogenic centre $\gamma^{\prime}$ to phosphorus. The relative configuration of the $\beta$-hydroxy phosphine oxide unit may not determine the geometry of the alkene products, but these stereogenic centres still act as temporary "relay" centres.

Determining the diastereomeric purity of compounds with remote stereogenic centres is a very difficult task indeed. ${ }^{26}$ Careful comparison of the $500 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectra of $\mathbf{5 6}$ and 59 allowed us to measure the diastereomeric purity of $\mathbf{5 9}$ confidently as $>95: 5$. We derivatised the diols $\mathbf{5 6}$ and $\mathbf{5 9}$ as ( $R$ )-MTPA ("Mosher's") diesters. The diol 59 was essentially diastereomerically pure ( $>95: 5$ ), so the ratio of diastereomeric esters obtained ( $\mathbf{7 0}: \mathbf{7 1}$ 89:11) reflected the enantiomeric purity of 59. The measured enantiomeric excess of 59 was $78 \%$, which

Table 2 Chemical shift differences in Mosher's diesters 70-73

|  |  |  | $\delta\left(\mathrm{H}^{\mathrm{x}}-\mathrm{H}^{\mathrm{Y}}\right)$ |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Entry | Mosher's <br> diester X | Mosher's <br> diester Y |  | $\mathrm{H}^{\mathrm{M}}$ | $\mathrm{H}^{\mathrm{N}}$ | $\mathrm{H}^{\mathrm{o}, \mathrm{O}^{\prime}}$ |
| 1 | $\mathbf{7 0}$ | $\mathbf{7 1}$ | -0.08 | -0.09 | -0.03 |  |
| 2 | $\mathbf{7 2}$ | $\mathbf{7 0}$ | +0.08 | +0.08 | +0.04 |  |

was the same (within experimental error) as that of the precursor ${ }^{17} 74(84 \%$ ee). The absolute stereochemistry of the stereogenic centre at C-4 (and hence the relative stereochemistry) of 59 was determined using Mosher's method ${ }^{27}$ (Table 2) $\ddagger$

The diastereomeric purity of $\mathbf{5 6}$ was measured by conversion into the corresponding Mosher's diesters; the ratio of peaks in the $500 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR spectrum of these Mosher's esters was $74: 26$ which reflects the ratio of $(4 R):(4 S)$ diols in the starting material (Table 2). ${ }^{27}$ The diol was known to have $84 \%$ ee. The 74:26 ratio of peaks observed does not correspond to the known enantiomeric excess (which would require a ratio of $92: 8$ ) so $\mathbf{5 6}$ must be contaminated with its diastereoisomer 59 (i.e. $\mathbf{7 2}+\mathbf{7 1 : 7 0}+\mathbf{7 3}$ was $74: 26$ ). We estimate that the diastereomeric purity of $\mathbf{5 6}$ was $80: 20$.

## Summary

We have developed stereoselective methods which allow complete control over the stereochemistry of the products. In the preceding paper, ${ }^{1}$ we described asymmetric syntheses of all four diastereomeric phosphine oxides $1\left(\mathrm{R}^{2}=\right.$ aryl). The work described in this paper allowed us to transform some aromatic $\mathrm{R}^{2}$ groups in phosphine oxides $\mathbf{1}$ into prochiral units, by oxidative cleavage of either an anisyl ring or a furan ring, which were

[^3]


66



67



Scheme 11
suitable for further functionalisation. Stereoselective reduction of the intermediate ketones gave the triols 55,58 and $\mathbf{6 8}$ with four controlled stereogenic centres. Horner-Wittig elimination of these triols was $E$-selective, yielding the diols 56 and 59 with 1,5 -related stereogenic centres across an $E$ alkene. This is the most remote chiral relationship which has been controlled using
phosphine oxide chemistry. Our work is neatly complemented by some methodology which has been reported by Thomas; transmetallation of chiral allylic stannanes such as 75, and reaction with aldehydes, gives homoallylic alcohols (e.g. 76) with 1,5 -related stereogenic centres across a $(Z)$-double bond. ${ }^{29}$

## Experimental

General methods have been described previously. ${ }^{1}$ Carbon NMR spectra were recorded with broad band proton decoupling and Attached Proton Test. Plus (+) and minus ( - ) symbols after the carbon NMR chemical shift indicate odd and even numbers of attached protons respectively.
( $2 R^{*}, 3 S^{*}, 4 S^{*}$ )-4-Diphenylphosphinoyl-1,2-epoxy-2-methyl-octan-3-ol 6
$m$-Chloroperbenzoic acid ( $57-85 \%$ by weight, $1.94 \mathrm{~g}, c a .7 .9$ mmol ) was added over 20 min to a stirred solution of ( $3 R^{*}$, $4 S^{*}$ )-4-diphenylphosphinoyl-2-methyloct-1-en-3-ol ${ }^{1} 5$ ( 1.00 g , 2.8 mmol ) and disodium hydrogen phosphate ( $2.16 \mathrm{~g}, 15.2$ $\mathrm{mmol})$ in dry dichloromethane $\left(40 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 16 h at $0^{\circ} \mathrm{C}$, quenched with sodium iodide ( $1.2 \mathrm{~g}, 7.9 \mathrm{mmol}$ ) and sodium thiosulfate $(1.1 \mathrm{~g}, 7.9$ mmol ) and extracted with dichloromethane ( $3 \times 30 \mathrm{~cm}^{3}$ ). The combined organic extracts were washed with saturated sodium bicarbonate solution $\left(30 \mathrm{~cm}^{3}\right)$ and brine $\left(30 \mathrm{~cm}^{3}\right)$ and evaporated under reduced pressure to give the epoxide $\mathbf{6}(1.06 \mathrm{~g}$, $>98 \%, 95: 5$ ratio of diastereomers) as an oil, $R_{\mathrm{f}} 0.28$ (EtOAc) (Found: $\mathrm{MH}^{+}, 359.1799 . \mathrm{C}_{21} \mathrm{H}_{27} \mathrm{O}_{3} \mathrm{P}$ requires $M H, 359.1776$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3383(\mathrm{br} \mathrm{s}, \mathrm{OH}), 1438(\mathrm{P}-\mathrm{Ph})$ and 1161 $(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.9-7.75(4 \mathrm{H}, \mathrm{m}), 7.6-7.25(6 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 4.07(1 \mathrm{H}, \mathrm{d}, 11.4, \mathrm{CHOH}), 3.70(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $2.99\left(1 \mathrm{H}, \mathrm{d}, J 5.1, \mathrm{C}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{O}\right), 2.61\left(1 \mathrm{H}, \mathrm{d}, J 5.1, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{O}\right)$, $2.44(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 1.89(1 \mathrm{H}, \mathrm{m}), 1.62(1 \mathrm{H}, \mathrm{m}), 1.31(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Me}), 1.3-0.9(4 \mathrm{H}, \mathrm{m})$ and $0.66\left(3 \mathrm{H}, \mathrm{t}, J 7.2\right.$, Me); $\delta_{\mathrm{C}}(100 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 132-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 69.8^{+}(\mathrm{CHOH}), 57.9^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}}\right.$ $17.3, \mathrm{MeC}), 51.3^{-}\left(\mathrm{CH}_{2} \mathrm{O}\right), 40.2^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.8, \mathrm{PCH}\right), 32.1^{-}(\mathrm{d}$, $\left.{ }^{2} J_{\mathrm{PC}} 6.6\right), 22.8^{-}, 21.6^{-}, 18.7^{+}(\mathrm{Me})$ and $13.6^{+}(\mathrm{Me}) ; m / z(\mathrm{FAB})$ $359.2\left(60 \%, \mathrm{MH}^{+}\right), 229.1$ (80), 202.1 ( $100, \mathrm{Ph}_{2} \mathrm{POH}$ ) and 201.1 ( $85, \mathrm{Ph}_{2} \mathrm{POH}$ ).

## $\left(2 R^{*}, 3 S^{*}, 4 R^{*}\right)$-6-Benzyloxy-4-diphenylphosphinoyl-1,2-epoxy-

 2-methylhexan-3-ol 9By the same general method, the allylic alcohol ${ }^{1} 8(547 \mathrm{mg}$, 1.36 mmol ) gave the epoxide $\mathbf{9}(534 \mathrm{mg}, 94 \%, 89: 11$ ratio of
diastereomers) as an oil, $R_{\mathrm{f}} 0.28$ (EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{C}_{3} \mathrm{H}_{5} \mathrm{O}$, 379.1464. $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{O}_{3} \mathrm{P}$ requires $\left.M-C_{3} H_{5} \mathrm{O}, 379.1463\right)$; $v_{\text {max }} /$ $\mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3389(\mathrm{br} \mathrm{s}, \mathrm{OH}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1160(\mathrm{P}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.0-7.75(4 \mathrm{H}, \mathrm{m}), 7.5-7.22(6 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{Ph}_{2} \mathrm{PO}\right), 4.34\left(1 \mathrm{H}, \mathrm{d}, J 11.9, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.26(1 \mathrm{H}, \mathrm{d}, J 11.9$, $\left.\mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.12(1 \mathrm{H}, \mathrm{d}, J 11.2, \mathrm{C} H \mathrm{OH}), 3.81(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $3.35\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OBn}\right), 3.13\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OBn}\right), 3.05$ $\left(1 \mathrm{H}, \mathrm{d}, J 5.1, \mathrm{C}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{O}\right), 2.88\left(1 \mathrm{H}, \mathrm{dt}, 5.5\right.$ and $\left.^{3} J_{\mathrm{PH}} 9.4, \mathrm{PCH}\right)$, $2.54\left(1 \mathrm{H}, \mathrm{d}, J 5.1, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{O}\right), 2.3-2.0(2 \mathrm{H}, \mathrm{m})$ and $1.27(3 \mathrm{H}$, $\mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(63 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $138.2^{-}$(ipso-Ph), 132-127 (m, $\left.\mathrm{Ph}_{2} \mathrm{PO}\right), 72.8^{-}\left(\mathrm{OCH} \mathrm{O}_{2} \mathrm{Ph}\right), 69.3^{+}(\mathrm{CHOH}), 67.8^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 6.8\right.$, $\left.\mathrm{PhCH} \mathrm{H}_{2} \mathrm{O}\right), 57.9^{-}\left(\mathrm{d},{ }^{3} \mathrm{JPC}_{\mathrm{PC}} 17.6, \mathrm{MeCO}\right), 50.8^{-}\left(\mathrm{CH}_{2} \mathrm{O}\right), 35.7^{+}(\mathrm{d}$, $\left.{ }^{1} J_{\mathrm{PC}} 69.9, \mathrm{PCH}\right), 22.8^{-}$and $18.5^{+}(\mathrm{Me}) ; m / z 379.1463(20 \%$, $\mathrm{M}^{+}-\mathrm{C}_{3} \mathrm{H}_{5} \mathrm{O}$ ), 271.1 (80) and 201.0 ( $100, \mathrm{Ph}_{2} \mathrm{PO}$ ).

## $\left(2 R^{*}, 3 R^{*}, 4 S^{*}\right)$-4-Diphenylphosphinoyl-2-methyl-1-phenyl-sulfanyloctan-2,3-diol 7

$n$-Butyllithium ( $0.26 \mathrm{~cm}^{3}$ of a $1.3 \mathrm{~mol} \mathrm{dm}^{-3}$ solution in hexanes, 0.33 mmol ) was added dropwise to a stirred solution of benzenethiol ( $37 \mu \mathrm{l}, 0.36 \mathrm{mmol}$ ) in dry THF ( $10 \mathrm{~cm}^{3}$ ). A solution of the phosphine oxide $6(109 \mathrm{mg}, 0.30 \mathrm{mmol})$ in dry THF $\left(5 \mathrm{~cm}^{3}\right)$ was added dropwise by cannula to the reaction mixture which was stirred for 30 min , quenched with saturated ammonium chloride solution ( $15 \mathrm{~cm}^{3}$ ) and extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ). The combined organic extracts were washed with saturated sodium bicarbonate solution ( $30 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $2: 1 \mathrm{EtOAc}$-hexane to give the diol $7(52 \mathrm{mg}, 37 \%)$ as minute needles, $\mathrm{mp} 146-147^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.32$ (3:2 EtOAc-hexane) (Found: C, 68.7; $\mathrm{H}, 7.05 ; \mathrm{P}, 6.8 \% ; \mathrm{MH}^{+}$, 469.1978. $\mathrm{C}_{27} \mathrm{H}_{33} \mathrm{O}_{3} \mathrm{PS}$ requires $\mathrm{C}, 69.2$; $\mathrm{H}, 7.10 ; \mathrm{P}, 6.6 \% ; M H, 469.1966) ; v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3346$ (br $\mathrm{s}, \mathrm{OH}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1159(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $8.0-7.1\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and PhS$), 4.02\left(1 \mathrm{H}, \mathrm{br} \mathrm{d},{ }^{3} J_{\mathrm{PH}} 14.5\right.$, $\mathrm{CHOH}), 3.89(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.37\left(1 \mathrm{H}, \mathrm{d}, J 13.4, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{SPh}\right)$, $3.05\left(1 \mathrm{H}, \mathrm{d}, J 13.4, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{SPh}\right), 2.89(1 \mathrm{H}, \mathrm{s}, \mathrm{OH}), 2.69(1 \mathrm{H}$, $\mathrm{m}, \mathrm{PCH}), 2.5-0.9(6 \mathrm{H}, \mathrm{m}), 1.27(3 \mathrm{H}, \mathrm{s}, \mathrm{Me})$ and $0.58(3 \mathrm{H}, \mathrm{t}$, $J 7.2, \mathrm{Me}$ ); $\delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 155.9^{+}$(ipso-Ph), 137-126 $\left(\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 74.8^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 13.3, \mathrm{CMeOH}\right), 72.9^{+}(\mathrm{CHOH})$, $44.6^{-}\left(\mathrm{PhSCH}_{2}\right), 37.4^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.8, \mathrm{PCH}\right), 32.4^{-}$and $22.8^{+}$ (Me); $m / z$ (FAB) 469.2 ( $80 \% \mathrm{MH}^{+}$).

## ( $2 R^{*}, 3 R^{*}, 4 S^{*}$ )-6-Benzyloxy-4-diphenylphosphinoyl-2-methyl-1-phenylsulfanylhexane-2,3-diol 10

By the same general method, the epoxide $9(480 \mathrm{mg}, 1.15 \mathrm{mmol})$ gave a crude product which was purified by flash chromatography eluting with EtOAc, to give the diol $\mathbf{1 0}(465 \mathrm{mg}, 77 \%$, $72 \%$ from 8) as plates, $\mathrm{mp} 141-142{ }^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.32$ (3:2 EtOAc-hexane) (Found: C, $69.9 ; \mathrm{H}, 6.40 ;$ P, $5.8 \%$; $\mathrm{MH}^{+}$, 547.2076. $\mathrm{C}_{32} \mathrm{H}_{35} \mathrm{O}_{4} \mathrm{PS}$ requires C, 70.3; H, 6.45; P, 5.7\%; MH, 547.2072); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3383$ (br s, OH), 1438 ( $\mathrm{P}-\mathrm{Ph}$ ) and $1160(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.85-7.1(20 \mathrm{H}$, $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and PhS$), 4.40\left(1 \mathrm{H}, \mathrm{d}, J 12.0, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.21(1 \mathrm{H}$, d, $\left.J 12.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.08(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.06\left(1 \mathrm{H}, \mathrm{br} \mathrm{d},{ }^{3} J_{\mathrm{PH}}\right.$ 14.0, CHOH ), $3.34(1 \mathrm{H}$, br s, OH$), 3.30\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OBn}\right)$, $3.27\left(1 \mathrm{H}, \mathrm{d}, J 13.0, \mathrm{C}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{SPh}\right), 3.17(1 \mathrm{H}, \mathrm{d}, J 12.9$, $\left.\mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{SPh}\right), 2.98\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OBn}\right), 2.87(1 \mathrm{H}, \mathrm{dt}, J 4.4$ and 9.5, PCH ), $2.49(1 \mathrm{H}, \mathrm{m})$ and $1.89(1 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(50 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $137.8^{-}$(ipso- Ph ), $136.9^{-}$(ipso- Ph ), 132-125 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ ), $75.0^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 12.1, \mathrm{CH}_{2} \mathrm{OBn}\right), 72.6^{-}\left(\mathrm{PhCH}_{2}\right), 72.9^{+}(\mathrm{CHOH})$, $72.3^{+}(\mathrm{CHOH}), 68.3^{-}(\mathrm{CMeOH}), 43.7^{-}\left(\mathrm{PhSCH}_{2}\right), 34.1^{+}$ (d, ${ }^{1} J_{\mathrm{PC}} 69.0, \mathrm{PCH}$ ), $23.1^{+}(\mathrm{Me})$ and $22.3^{-} ; \mathrm{m} / \mathrm{z}$ (FAB) 547.2 $\left(70 \%, \mathrm{MH}^{+}\right), 201.0\left(80, \mathrm{Ph}_{2} \mathrm{PO}\right)$ and $91.1(100, \mathrm{Bn})$.

## ( $\boldsymbol{Z}$ )-6-Benzyloxy-2-methyl-1-phenylsulfanylhex-3-en-2-ol 11

Sodium hydride ( 16 mg of a $60 \%$ dispersion in oil, 0.40 mmol ) was added to a stirred solution of the diol $\mathbf{1 0}(51 \mathrm{mg}, 96 \mu \mathrm{~mol})$
in dry DMF $\left(2 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$. The reaction was stirred at $60^{\circ} \mathrm{C}$ for 1 h , cooled to $20^{\circ} \mathrm{C}$, diluted with saturated brine $\left(10 \mathrm{~cm}^{3}\right)$ and extracted into $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 10 \mathrm{~cm}^{3}\right)$. The combined organic extracts were washed with water $\left(10 \mathrm{~cm}^{3}\right)$ and saturated brine $\left(10 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $4: 1$ hexane-EtOAc to give the alkene $11(4.1 \mathrm{mg}, 14 \%)$ as an oil, $R_{\mathrm{f}} 0.50(3: 1$ hexane$\mathrm{EtOAc}) ; v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3500(\mathrm{br} \mathrm{s}, \mathrm{OH}), 1630(\mathrm{C}=\mathrm{C}), 1437$ $(\mathrm{P}-\mathrm{Ph})$ and $1160(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.4-7.1(10 \mathrm{H}$, $\mathrm{m}, \mathrm{Ph}$ and PhS ), $5.59\left(1 \mathrm{H}\right.$, br d, $\left.J 12.1, \mathrm{C} H=\mathrm{CHCH}_{2}\right), 5.46$ $\left(1 \mathrm{H}, \mathrm{td}, J 8.2\right.$ and $\left.11.8, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.53(2 \mathrm{H}, \mathrm{AB}$ q, $J 12.1$, $\left.\mathrm{PhCH}_{2}\right), 3.81(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.50\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OBn}\right), 3.17$ $\left(1 \mathrm{H}, \mathrm{d}, J 12.7, \mathrm{PhSCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.09\left(1 \mathrm{H}, \mathrm{d}, J 12.7, \mathrm{PhSCH}_{\mathrm{A}} H_{\mathrm{B}}\right)$, $2.74(1 \mathrm{H}, \mathrm{m}), 2.59(1 \mathrm{H}, \mathrm{m})$ and $1.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 136.9^{+}(\mathrm{CH}=\mathrm{CH}), 129-127(\mathrm{~m}, \mathrm{Ph}$ and PhS$), 125.9^{+}$ $(\mathrm{CH}=\mathrm{CH}), 73.7^{+}(\mathrm{MeCOH}), 73.2^{-}$, 68.9-9, 48.3- $\left(\mathrm{PhSCH}_{2}\right)$, $28.9^{-}$and $28.4^{+}(\mathrm{Me})$.

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)hexyl acetate 27

$\left(1 R^{*}, 2 S^{*}\right)$-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)-hexan-1-ol ${ }^{1} \mathbf{1 6}(1.245 \mathrm{~g}, 3.05 \mathrm{mmol})$ was dissolved in pyridine $\left(10 \mathrm{~cm}^{3}\right)$ and acetic anhydride $\left(10 \mathrm{~cm}^{3}\right)$, and the reaction mixture was stirred for 3 days. The reaction was diluted with ethyl acetate $\left(30 \mathrm{~cm}^{3}\right)$, washed with hydrochloric acid $(2 \times 30$ $\mathrm{cm}^{3}$ ), saturated aqueous sodium bicarbonate solution $\left(30 \mathrm{~cm}^{3}\right)$, saturated brine ( $30 \mathrm{~cm}^{3}$ ) and saturated aqueous copper(II) nitrate solution $\left(30 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product. Purification by flash chromatography eluting with EtOAc gave the ester 27 ( 992 mg , $72 \%$ ) as needles, $\mathrm{mp} 185-186^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.38$ (EtOAc) (Found: C, 72.3; H, 7.00; P, 6.9\%; M ${ }^{+}$, 450.1952. $\mathrm{C}_{27} \mathrm{H}_{31} \mathrm{PO}_{4}$ requires C, 72.0; H, 6.95; P, $6.9 \% ; M, 450.1960$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1741(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $7.95-7.25\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 7.13(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar})$, $6.70(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.20(1 \mathrm{H}, \mathrm{dd}, J 4.7$ and $8.7, \mathrm{CHOAc})$, $3.68(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.73\left(1 \mathrm{H}, \mathrm{qd}, J 5.0\right.$ and $\left.{ }^{3} J_{\mathrm{PH}} 9.9, \mathrm{PCH}\right)$, $2.0-0.85(9 \mathrm{H}, \mathrm{m})$ and $0.56(3 \mathrm{H}, \mathrm{t}, J 6.7, \mathrm{Me}) ; \delta_{\mathrm{C}}(50 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $169.2^{-}(\mathrm{C}=\mathrm{O}), 158.9^{-}$(ipso-Ar), 133-127 (m, Ph ${ }_{2} \mathrm{PO}$ and remaining Ph), $113.5^{+}(\mathrm{Ar}), 72.4^{+}(\mathrm{CHOAc}), 55.0^{+}(\mathrm{OMe})$, $44.9^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 67.8, \mathrm{PCH}\right), 31.0^{-}$(d, ${ }^{2} J_{\mathrm{PC}} 6.8$ ), $23.7^{-}, 22.3^{-}$, $20.7^{+}(\mathrm{OAc})$ and $13.2^{+}(\mathrm{Me}) ; m / z 450.1\left(5 \%, \mathrm{M}^{+}\right)$and 202.1 (100, $\mathrm{Ph}_{2} \mathrm{POH}$ ).

## ( $1 R^{*}, 2 R^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)hexyl acetate 30

By the same general method, ( $1 R^{*}, 2 R^{*}$ )-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexan-1-ol ${ }^{1} \quad \mathbf{2 0} \quad\left(\begin{array}{llll}2.20 & \mathrm{~g}, & 5.39\end{array}\right.$ mmol ), pyridine ( $17.5 \mathrm{~cm}^{3}$ ) and acetic anhydride ( $17.5 \mathrm{~cm}^{3}$ ) gave a crude product after 2 days. Purification by flash chromatography eluting with EtOAc gave the ester $30(1.81 \mathrm{~g}, 75 \%)$ as an oil, $R_{\mathrm{f}} 0.47$ (EtOAc) (Found: $\mathrm{M}^{+}, 450.1954 . \mathrm{C}_{27} \mathrm{H}_{31} \mathrm{PO}_{4}$ requires $M, 450.1960) ; v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1738(\mathrm{C}=\mathrm{O}), 1438$ ( $\mathrm{P}-\mathrm{Ph}$ ) and $1178(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.0-7.4(10 \mathrm{H}$, $\left.\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 7.29$ ( $2 \mathrm{H}, \mathrm{d}, J 8.6, \mathrm{Ar}$ ), 6.83 ( $2 \mathrm{H}, \mathrm{d}, J 8.6, \mathrm{Ar}$ ), 6.03 $(1 \mathrm{H}, \mathrm{dd}, J 8.3$ and $9.9, \mathrm{CHOAc}), 3.74(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.96(1 \mathrm{H}$, $\mathrm{qd}, J 4.8$ and $\left.^{3} J_{\mathrm{PH}} 9.6, \mathrm{PCH}\right), 1.6-0.8(6 \mathrm{H}, \mathrm{m}), 1.3(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ and $0.47(3 \mathrm{H}, \mathrm{t}, J 6.7, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 168.2^{-}(\mathrm{C}=\mathrm{O})$, $159.3^{-}$(ipso-Ar), $135-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ph$)$, $113.6^{+}(\mathrm{Ar}), 74.3^{+}(\mathrm{CHOAc}), 55.1^{+}(\mathrm{OMe}), 42.6^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.5\right.$, PCH), 29.5- (d, $\left.{ }^{2} J_{\mathrm{PC}} 6.0\right), 25.2^{-}, 22.3^{-}, 20.0^{+}(\mathrm{OAc})$ and $13.1^{+}$ $(\mathrm{Me}) ; m / z 450.1\left(5 \%, \mathrm{M}^{+}\right)$and $202.1\left(100, \mathrm{Ph}_{2} \mathrm{POH}\right)$.

## ( $1 R^{*}, 2 S^{*}$ )-4-Benzyloxy-2-diphenylphosphinoyl-1-(4-methoxyphenyl)butyl acetate 34

By the same general method, $\left(1 R^{*}, 2 S^{*}\right)$-4-benxyloxy-2-di-phenylphosphinoyl-1-(4-methoxyphenyl)butan-1-ol ${ }^{1} 13$ (3.48
g, 7.16 mmol ), pyridine $\left(25 \mathrm{~cm}^{3}\right)$ and acetic anhydride $\left(25 \mathrm{~cm}^{3}\right)$ gave a crude product after 1 day. Purification by flash chromatography eluting with EtOAc gave the ester $34(3.37 \mathrm{~g}, 89 \%)$ as an oil, $R_{\mathrm{f}} 0.33$ (EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{Ac}, 485.1882 . \mathrm{C}_{32} \mathrm{H}_{33} \mathrm{PO}_{5}$ requires $M-A c, 485.1882)$; $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1741(\mathrm{C}=\mathrm{O})$; $1438(\mathrm{P}-\mathrm{Ph})$ and 1193 ( $\mathrm{P}=\mathrm{O}$ ); $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) 8.0-7.25 ( $17 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}, \mathrm{Ph}$ and remaining Ar), 6.77 ( $2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}$ ), $6.23(1 \mathrm{H}, \mathrm{dd}, J 3.3$ and $9.1, \mathrm{CHOAc}), 4.01(2 \mathrm{H}, \mathrm{AB}$ q,$J 12.9$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 3.73 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), $3.2-2.9(3 \mathrm{H}, \mathrm{m}, \mathrm{PCH}$ and $\mathrm{CH}_{2} \mathrm{OBn}$ ) and $2.25(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 169.2^{-}$ (C=O), 158.9 ${ }^{-}$, $155.7^{-}$(ipso- Ar and Ph ), 133-126 (m, Ph ${ }_{2} \mathrm{PO}$ and remaining Ph$), 113.6^{+}(\mathrm{Ar}), 72.2^{-}\left(\mathrm{OCH}_{2} \mathrm{Ph}\right), 71.7^{+}$ (CHOAc), $67.9^{-}\left(\mathrm{CH}_{2} \mathrm{OBn}\right), 55.1^{+}(\mathrm{OMe}), 40.5^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.1\right.$, PCH ), $34.0^{-}$and $20.7^{+}(\mathrm{OAc}) ; \mathrm{m} / \mathrm{z} 485.2$ ( $5 \% \mathrm{M}^{+}-\mathrm{Ac}$ ), 202.1 (100, $\left.\mathrm{Ph}_{2} \mathrm{POH}\right), 201.0\left(75, \mathrm{Ph}_{2} \mathrm{POH}\right)$ and 91.1 ( $90, \mathrm{Bn}$ ).

## (1S,2R,4R)-4-Benzyloxy-2-diphenylphosphinoyl-1-(4-methoxyphenyl)octyl acetate 42

By the same general method, the phosphine oxide 41 ( 251 mg , 0.46 mmol ), pyridine ( $2.5 \mathrm{~cm}^{3}$ ) and acetic anhydride ( $2.5 \mathrm{~cm}^{3}$ ) gave a crude product after 2 days. Purification by flash chromatography eluting with EtOAc gave the ester 42 ( $234 \mathrm{mg}, 87 \%$ ) as an oil, $R_{\mathrm{f}} 0.51$ (EtOAc); $[a]_{\mathrm{D}}^{20}+2.8\left(c 0.91\right.$ in $\mathrm{CHCl}_{3}$ ) (Found: $\mathrm{M}^{+}$, 584.2691. $\mathrm{C}_{36} \mathrm{H}_{41} \mathrm{PO}_{5}$ requires $M, 584.2693$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 1740(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1192(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(200$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 7.9-7.2 ( $15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), 7.13 ( $2 \mathrm{H}, \mathrm{d}$, $J 8.7, \mathrm{Ar}), 6.76(2 \mathrm{H}, \mathrm{d}, J 8.8, \mathrm{Ar}), 6.15(1 \mathrm{H}, \mathrm{dd}, J 2.9$ and 10.5 , CHOAc), 4.22 ( $1 \mathrm{H}, \mathrm{d}, J 12.0, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}$ ), 3.91 ( $1 \mathrm{H}, \mathrm{d}, J 12.0$, $\mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}$ ), $3.75(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.94(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{OBn}), 2.73$ ( 1 H , quin, J 5.9, PCH), $2.1(2 \mathrm{H}, \mathrm{m}), 1.98(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}), 1.3-0.9$ $(6 \mathrm{H}, \mathrm{m})$ and $0.78(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $169.4^{-}$(C=O), $159.0^{-}$(ipso-Ar), 139.1- (ipso-Ar), 134-127 (m, $\mathrm{Ph}_{2} \mathrm{PO}, \mathrm{Ph}$ and remaining Ar ), $113.6^{+}(\mathrm{Ar}), 76.7^{+}$(d, J 4.0, $\mathrm{CHOBn}), 72.7^{+}(\mathrm{CHOAc}), 55.2^{+}(\mathrm{OMe}), 41.0^{+}\left(\mathrm{d},{ }^{1} \mathrm{~J}_{\mathrm{PC}} 67.8\right.$, PCH ), 32.9-, 28.2-, 26.6 ${ }^{-}$, 22.6 ${ }^{-}$, 20.9 $9^{+}$(OAc) and 13.9+ (Me); $\mathrm{m} / \mathrm{z} 587.1\left(45 \%, \mathrm{M}^{+}\right), 541.3(80, \mathrm{M}-\mathrm{Ac})$ and 232.1 (100).

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)hexyl benzoate 17

Triethylamine ( $1.78 \mathrm{~g}, 17.4 \mathrm{mmol}$ ) and benzoyl chloride ( 2.17 g , $15.3 \mathrm{mmol})$ were added dropwise to a solution of $\left(1 R^{*}, 2 S^{*}\right)-2$ -diphenylphosphinoyl-1-(4-methoxyphenyl)hexan-1-ol 16 (1.245 $\mathrm{g}, 3.05 \mathrm{mmol}$ ) and $N, N$-dimethylaminopyridine ( $99 \mathrm{mg}, 0.81$ mmol ) in dry dichloromethane ( $20 \mathrm{~cm}^{3}$ ) at room temperature. The reaction was stirred for 3 days, quenched with water ( $20 \mathrm{~cm}^{3}$ ), extracted with dichloromethane ( $3 \times 20 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product. Purification by flash chromatography eluting with $2: 1$ EtOAc-hexane gave the ester $17(1.51 \mathrm{~g}, 95 \%)$ as needles, $\mathrm{mp} 144-145^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.51$ (EtOAc) (Found: C, 74.4; H, 6.70; P, 6.0\%; M ${ }^{+}$, 512.2114. $\mathrm{C}_{32} \mathrm{H}_{33} \mathrm{PO}_{4}$ requires $\mathrm{C}, 74.4 ; \mathrm{H}, 6.65 ; \mathrm{P}, 6.2 \% ; M, 512.2109) ; v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 1721(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1212(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(200$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.01(2 \mathrm{H}, \mathrm{dd}, J 1.3$ and 7.2, ortho-Bz), $7.9-7.3$ $\left(13 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Bz$)$, $7.13(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar})$, 6.71 ( $2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.32(1 \mathrm{H}, \mathrm{dd}, J 4.8$ and $8.5, \mathrm{CHOBz})$, $3.71(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.88\left(1 \mathrm{H}, \mathrm{qd}, J 5.0\right.$ and $\left.{ }^{2} J_{\mathrm{PH}} 10.1, \mathrm{PCH}\right)$, 2.1-1.8 ( $2 \mathrm{H}, \mathrm{m}$ ), $1.05(4 \mathrm{H}, \mathrm{m})$ and $0.62(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me})$; $\delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 169.2^{-}(\mathrm{C}=\mathrm{O}), 159.1^{-}$(ipso-Ar), 134-127 $\left(\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ar and Bz ), $113.7^{+}$(Ar), $73.8^{+}$ $(\mathrm{CHOBz}), 55.2^{+}(\mathrm{OMe}), 45.2^{+}\left(\mathrm{d},{ }^{1} \mathrm{~J}_{\mathrm{PC}} 67.3, \mathrm{PCH}\right), 31.0^{-}$ (d, ${ }^{2} J_{\mathrm{PC}} 6.2$ ), $24.1^{-}, 22.6^{-}$and $13.5^{+}(\mathrm{Me}) ; ~ m / z 512.2\left(10 \%, \mathrm{M}^{+}\right)$ and 105 ( $100, \mathrm{PhCO}$ ).

## ( $1 R^{*}, 2 R^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)hexyl benzoate 21

By the same general method, $\left(1 R^{*}, 2 R^{*}\right)$-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexan-1-ol ${ }^{1} \quad \mathbf{2 0} \quad(2.93 \mathrm{~g}, 7.18$
mmol ) gave a crude product after 2 days. Purification by flash chromatography eluting with $2: 1$ EtOAc-hexane gave the ester $21(2.97 \mathrm{~g}, 81 \%)$ as needles, mp 181-183 ${ }^{\circ} \mathrm{C}$ (from EtOAchexane); $R_{\mathrm{f}} 0.53$ (EtOAc) (Found: C, 74.4; H, 6.45; P, 6.1\%; $\mathrm{M}^{+}$, 512.2116. $\mathrm{C}_{32} \mathrm{H}_{33} \mathrm{PO}_{4}$ requires $\mathrm{C}, 74.4 ; \mathrm{H}, 6.65 ; \mathrm{P}, 6.2 \% ; M$, 512.2109); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1719(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.12(2 \mathrm{H}$, dd, $J 1.3$ and 8.5 , ortho- Bz ), $8.0-7.3\left(13 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Bz), $7.18(2 \mathrm{H}, \mathrm{d}, J 8.7$, Ar), 6.82 ( $2 \mathrm{H}, \mathrm{d}, J$ 8.7, Ar), $6.38(1 \mathrm{H}, \mathrm{t}, J 9.0, \mathrm{CHOBz}$ ), 3.86 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), $3.13(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 1.7-0.8(6 \mathrm{H}, \mathrm{m})$ and 0.54 ( $3 \mathrm{H}, \mathrm{t}, J 7.4, \mathrm{Me}$ ); $\delta_{\mathrm{C}}\left(63 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 164.9^{-}$(C=O), $159.4^{-}$ (ipso-Ar), $135-126\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ar and Bz$)$, $113.8^{+}(\mathrm{Ar}), 75.2^{+}(\mathrm{CHOBz}), 55.2^{+}(\mathrm{OMe}), 43.3^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.1\right.$, PCH), $29.9^{-}$(d, ${ }^{2} J_{\text {PC }} 5.9$ ), $25.6^{-}, 24.6^{-}$and $13.4^{+}(\mathrm{Me}) ; \mathrm{m} / \mathrm{z}$ $512.2\left(75 \%, \mathrm{M}^{+}\right), 202.1\left(85, \mathrm{Ph}_{2} \mathrm{POH}\right)$ and $105(100, \mathrm{PhCO})$.

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)propyl benzoate 25

By the same general method, ( $1 R^{*}, 2 S^{*}$ )-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)propan-1-ol ${ }^{1} \mathbf{2 4}(451 \mathrm{mg}, 31.23$ mmol ) gave a crude product after 2 days. Purification by flash chromatography eluting with 2:1 EtOAc-hexane gave the ester $25(1.51 \mathrm{~g},>98 \%)$ as needles, $\mathrm{mp} 165-166^{\circ} \mathrm{C}$ (from EtOAchexane); $R_{\mathrm{f}} 0.51$ (EtOAc) (Found: C, 73.5; H, 5.60; P, 6.6\%; $\mathrm{M}^{+}$, 470.1635. $\mathrm{C}_{29} \mathrm{H}_{27} \mathrm{PO}_{4}$ requires $\mathrm{C}, 74.0 ; \mathrm{H}, 5.80 ; \mathrm{P}, 6.6 \% ; M$, 470.1647); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1723(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph})$; $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.95(2 \mathrm{H}, \mathrm{dd}, J 1.3$ and 7.2 , ortho-Bz), 7.9-7.3 ( $13 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and remaining Bz), $7.15(2 \mathrm{H}, \mathrm{d}, J 8.7$, Ar), $6.72(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.39(1 \mathrm{H}$, dd, $J 4.8$ and 8.5 , $\mathrm{C} H \mathrm{OBz}), 3.65(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.93\left(1 \mathrm{H}, \mathrm{qd}, J 5.0\right.$ and ${ }^{2} J_{\mathrm{PH}} 10.1$, $\mathrm{PCH})$ and $1.30\left(3 \mathrm{H}\right.$, dd, $J 7.3$ and $\left.{ }^{3} J_{\mathrm{PH}} 16.0, \mathrm{Me}\right) ; \delta_{\mathrm{C}}(50 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) 164.6- (C=O), $158.9^{-}$(ipso-Ar), 133-127 (m, Ph ${ }_{2} \mathrm{PO}$ and remaining Ar and Bz ), 113.6 ${ }^{+}$(Ar), 73.1+ (CHOBz), 54.9+ $(\mathrm{OMe}), 39.7^{+}$(d, $\left.{ }^{1} \mathrm{~J}_{\mathrm{PC}} 68.5, \mathrm{PCH}\right)$ and $8.3^{+}(\mathrm{Me}) ; \mathrm{m} / \mathrm{z} 470.2$ $\left(80 \%, \mathrm{M}^{+}\right)$, 365.1 ( $100, \mathrm{M}-\mathrm{PhCO}$ ) and $202.1\left(95, \mathrm{Ph}_{2} \mathrm{POH}\right)$.

## $\left(1 R^{*}, 2 S^{*}\right)$-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)butane-1,4-diyl bis(benzyl ether) 14

Benzyl bromide ( 0.75 mmol ) was added dropwise to a stirred solution of ( $1 R^{*}, 2 S^{*}$ )-4-benzyloxy-2-diphenylphosphinoyl-1-(4-methoxyphenyl)butan-1-ol ${ }^{1}$ ( $262 \mathrm{mg}, 0.54 \mathrm{mmol}$ ), sodium hydride ( $28 \mathrm{mg}, 60 \%$ dispersion in oil, 0.7 mmol ) and tetra- $n$ butylammonium iodide ( $5 \mathrm{mg}, 13 \mu \mathrm{~mol}$ ) in dry THF ( $10 \mathrm{~cm}^{3}$ ) at room temperature. The reaction was stirred for 3 days, quenched with water $\left(10 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane $\left(3 \times 10 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product. Purification by flash chromatography eluting with $1: 1 \mathrm{EtOAc}$-hexane, to give the dibenzyl ether 14 ( $322 \mathrm{mg},>98 \%$ ) as an oil, $R_{\mathrm{f}} 0.49$ ( EtOAc ) (Found: $\mathrm{M}^{+}-\mathrm{Bn}, 485.1878 . \mathrm{C}_{37} \mathrm{H}_{37} \mathrm{O}_{4} \mathrm{P}$ requires $M-B n, 485.1882$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1437(\mathrm{P}-\mathrm{Ph})$ and $1201(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(200 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.0-7.1\left(22 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}, 2 \times \mathrm{Ph}\right.$ and remaining Ar), $6.76(2 \mathrm{H}$, br d, $J 8.7, \mathrm{Ar}), 5.01\left(1 \mathrm{H}, \mathrm{dd}, J 4.3\right.$ and ${ }^{3} J_{\mathrm{PH}} 7.9$, $\mathrm{C} H \mathrm{OBn}), 4.37\left(1 \mathrm{H}, \mathrm{d}, J 11.3, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.18(1 \mathrm{H}, \mathrm{d}, J 11.3$, $\left.\mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.10\left(2 \mathrm{H}, \mathrm{AB} \mathrm{q}, J 11.8, \mathrm{PhCH}_{2}\right), 3.75(3 \mathrm{H}$, s, OMe), 3.3-2.9 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OBn}$ ) and 2.3-2.1 $(2 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 158.9^{-}, 155.9^{-}$, $138.4^{-}$(ipso-Ph), $138.0^{-}$ (ipso- Ph ), 134-127 ( $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar and $2 \times \mathrm{Ph}$ ), $113.6^{+}(\mathrm{Ar}), 77.9^{+}(\mathrm{CHOBn}), 72.3^{-}, 70.7^{-}, 68.6^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 17.0\right.$, $\mathrm{CH}_{2} \mathrm{OBn}$ ), $55.1^{+}(\mathrm{OMe}), 43.2^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.4, \mathrm{PCH}\right)$ and $24.5^{-}$; $m / z 485.2\left(45 \%, \mathrm{M}^{+}-\mathrm{Bn}\right)$ and $91.1(100, \mathrm{Bn})$.

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-4-hydroxy-1-(4-methoxyphenyl)butyl acetate 35

A solution of ( $1 R^{*}, 2 S^{*}$ )-4-benzyloxy-2-diphenylphosphinoyl-1-(4-methoxyphenyl)butyl acetate 34 ( $510 \mathrm{mg}, 0.97 \mathrm{mmol}$ ) and palladium on carbon ( $5 \%$ by weight, 48 mg ) in methanol $\left(20 \mathrm{~cm}^{3}\right)$ was degassed and flushed with argon, degassed and
flushed with hydrogen ( $\times 2$ ), and stirred for 3 days under a hydrogen atmosphere. The reaction mixture was filtered through Celite with dichloromethane ( $50 \mathrm{~cm}^{3}$ ) and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $5 \%$ methanol in EtOAc, to give the alcohol $35(330 \mathrm{mg}, 78 \%)$ as minute needles, mp 164-165 ${ }^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.43$ (EtOAc) (Found: C, 68.4; H, 7.25; P, 8.0\%; $\mathrm{M}^{+}$, 438.1601. $\mathrm{C}_{25} \mathrm{H}_{27} \mathrm{O}_{5} \mathrm{P}$ requires $\mathrm{C}, 68.4 ; \mathrm{H}, 7.20 ; \mathrm{P}, 8.0 \% ; M, 438.1596) ; v_{\max } / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 3379(\mathrm{OH}), 1743(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}(400$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $7.9-7.4\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 7.07(2 \mathrm{H}, \mathrm{d}, J 8.7$, Ar), $6.74(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.13\left(1 \mathrm{H}, \mathrm{dd}, J 3.9\right.$ and ${ }^{3} J_{\mathrm{PH}} 8.5$, CHOAc), $3.72(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $3.63(1 \mathrm{H}, \mathrm{m}), 3.48(1 \mathrm{H}, \mathrm{m}), 3.00$ $(1 \mathrm{H}, \mathrm{m}), 2.08(1 \mathrm{H}, \mathrm{m}), 2.04(1 \mathrm{H}, \mathrm{m})$ and $1.94(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 169.2^{-}(\mathrm{C}=\mathrm{O}), 159.2^{-}$(ipso-Ar), 132-127 $\left(\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ar$), 114.0^{+}(\mathrm{Ar}), 72.3^{+}(\mathrm{CHOAc})$, $60.4^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 5.0, \mathrm{CH}_{2} \mathrm{OH}\right), 55.2^{+}(\mathrm{OMe}), 42.7^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 66.8\right.$, $\mathrm{PCH}), 27.4^{-}$and $20.9^{+}(\mathrm{OAc}) ; m / z 438.2\left(10 \%, \mathrm{M}^{+}\right)$and 202.1 (100, $\left.\mathrm{Ph}_{2} \mathrm{POH}\right)$.

## (1S,2R,4R)-2-Diphenylphosphinoyl-4-hydroxy-1-(4-methoxyphenyl)octyl acetate 43

A solution of $(1 S, 2 R, 4 R)$-4-benzyloxy-2-diphenylphosphinoyl-1-(4-methoxyphenyl)octyl acetate $\mathbf{4 2}(191 \mathrm{mg}, 0.32 \mathrm{mmol})$ and palladium on carbon ( $5 \%$ by weight, 16 mg ) in methanol $\left(10 \mathrm{~cm}^{3}\right)$ was agitated under an atmosphere of hydrogen (4 atm) for 2 days, filtered through Celite with dichloromethane $\left(50 \mathrm{~cm}^{3}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with EtOAc to give the alcohol $43(84 \mathrm{mg}, 52 \%, 72 \%$ based on recovered starting material) as minute needles, $\mathrm{mp} 170-171^{\circ} \mathrm{C}$ (from EtOAc-hexane); $R_{\mathrm{f}} 0.22$ (EtOAc) (Found: C, 70.3; H, 7.25; P, 6.2\%; M ${ }^{+}-\mathrm{H}, 495.2295 . \mathrm{C}_{29} \mathrm{H}_{35} \mathrm{O}_{5} \mathrm{P}$ requires C, 70.4; $\mathrm{H}, 7.15 ; \mathrm{P}, 6.2 \% ; M-H, 495.2300) ; v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3359$ $(\mathrm{OH}), 1743(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1176(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 7.9-7.4\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 7.07(2 \mathrm{H}, \mathrm{d}, J 8.6, \mathrm{Ar}), 6.73$ ( $2 \mathrm{H}, \mathrm{d}, J 8.6, \mathrm{Ar}$ ), $6.12\left(1 \mathrm{H}, \mathrm{dd}, J 4.3\right.$ and $\left.{ }^{3} J_{\mathrm{PH}} 7.9, \mathrm{C} H O A c\right)$, $3.75(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.70(1 \mathrm{H}, \mathrm{m}), 3.56(1 \mathrm{H}, \mathrm{m}), 3.06(1 \mathrm{H}$, $\mathrm{m}, \mathrm{PCH}), 2.2-0.95(10 \mathrm{H}, \mathrm{m})$ and $0.78(3 \mathrm{H}, \mathrm{t}, J 7.3$, Me); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 169.2^{-}(\mathrm{C}=\mathrm{O}), 159.1^{-}$(ipso-Ar), 132-126 $\left(\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ar$), 113.9^{+}(\mathrm{Ar}), 72.8^{+}(\mathrm{CHOAc})$, $68.2^{+}(\mathrm{CHOH}), 55.2^{+}(\mathrm{OMe}), 42.3^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 66.4, \mathrm{PCH}\right), 37.3^{-}$, $31.9^{-}, 27.8^{-}, 22.5^{-}, 21.0^{+}(\mathrm{OAc})$ and $13.9^{+}(\mathrm{Me}) ; \mathrm{m} / \mathrm{z} 495.2$ $\left(10 \%, \mathrm{M}^{+}-\mathrm{H}\right), 435.2(100)$ and $201.1\left(60, \mathrm{Ph}_{2} \mathrm{PO}\right)$. Recovered starting material was also obtained ( $51 \mathrm{mg}, 27 \%$ ).

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)-4phthalimidobutyl acetate 36

Diethyl azodicarboxylate ( $0.71 \mathrm{~cm}^{3}, 4.52 \mathrm{mmol}$ ) was added dropwise to a solution of $\left(1 R^{*}, 2 S^{*}\right)$-2-diphenylphosphinoyl-4-hydroxy-1-(4-methoxyphenyl)butyl acetate $35(1.34 \mathrm{~g}, 3.06$ $\mathrm{mmol})$, triphenylphosphine ( $1.20 \mathrm{~g}, 4.58 \mathrm{mmol}$ ) and phthalimide ( $532 \mathrm{mg}, 3.64 \mathrm{mmol}$ ) in dry THF $\left(60 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$. The reaction was stirred for 16 h and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with 78:17:5 EtOAc-hexanemethanol and HPLC eluting with $0.8 \%$ methanol in chloroform, to give the phthalimide $36(1.26 \mathrm{~g}, 73 \%)$ as an oil, HPLC retention time $19 \mathrm{~min} ; R_{\mathrm{f}} 0.34$ (EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{Ac}$, 524.1617. $\mathrm{C}_{33} \mathrm{H}_{30} \mathrm{NO}_{6} \mathrm{P}$ requires $M-A c, 524.1627$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 1770$ (imide $\mathrm{C}=\mathrm{O}$ ), $1741(\mathrm{C}=\mathrm{O}), 1708$ (imide $\mathrm{C}=\mathrm{O}$ ) and 1438 (P-Ph); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.0-7.4\left(14 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar), $6.98(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.43(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.11$ $\left(1 \mathrm{H}, \mathrm{dd}, J 3.3\right.$ and $\left.{ }^{3} J_{\mathrm{PH}} 8.7, \mathrm{CHOAc}\right), 3.48(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.28$ $\left(1 \mathrm{H}, \mathrm{td}, J 5.0\right.$ and $\left.13.9, \mathrm{NCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.12(1 \mathrm{H}$, ddd, $J 5.0,8.9$ and $\left.13.9, \mathrm{NCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 2.90\left(1 \mathrm{H}, \mathrm{qd}, J 4.0\right.$ and $\left.{ }^{2} J_{\mathrm{PH}} 9.2, \mathrm{PCH}\right)$, $2.23(2 \mathrm{H}, \mathrm{m})$ and $1.99(3 \mathrm{H}, \mathrm{Ac}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 169.3^{-}$, $168.4^{-}(\mathrm{C}=\mathrm{O} \times 3)$, $158.7^{-}$(ipso-Ar), $155.8^{+}(\mathrm{Ar}), 133.5^{+}(\mathrm{Ar})$,

134-128 ( $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar ), $126.8^{+}$(Ar), $122.9^{+}$ (Ar), $113.6^{+}(\mathrm{Ar}), 71.7^{+}(\mathrm{CHOAc}), 54.8^{+}(\mathrm{OMe}), 42.4^{+}(\mathrm{d}$, ${ }^{1} J_{\mathrm{PC}} 67.0, \mathrm{PCH}$ ), $37.7^{-}$(d, ${ }^{3} J_{\mathrm{PC}} 5.2, \mathrm{CHN}$ ), $22.6^{-}$and $21.0^{+}$ $(\mathrm{OAc}) ; m / z 524.2\left(10 \%, \mathrm{M}^{+}-\mathrm{Ac}\right), 202.1\left(90, \mathrm{Ph}_{2} \mathrm{POH}\right)$ and 77.0 (100, Ph).

## (1S,2R,3E)-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)oct-3enyl acetate 44

Diethyl azodicarboxylate ( $0.29 \mathrm{~cm}^{3}, 1.82 \mathrm{mmol}$ ) was added dropwise to a solution of $(1 S, 2 R, 4 R)$-2-diphenylphosphinoyl-4-hydroxy-1-(4-methoxyphenyl)octyl acetate 43 ( $610 \mathrm{mg}, 1.24$ mmol ), triphenylphosphine ( $493 \mathrm{mg}, 1.88 \mathrm{mmol}$ ) and phthalimide ( $220 \mathrm{mg}, 1.50 \mathrm{mmol}$ ) in dry THF $\left(30 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$. The reaction was stirred for 16 h and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $1: 1 \mathrm{EtOAc}$-hexane and then EtOAc to give the allylic phosphine oxide $44(373 \mathrm{mg}, 63 \%)$ as an oil, $R_{\mathrm{f}} 0.29$ ( EtOAc ); $[a]_{\mathrm{D}}^{20}-1.7$ (c 0.91 in $\mathrm{CHCl}_{3}$ ) (Found: $\mathrm{M}^{+}$, 476.2112. $\mathrm{C}_{29} \mathrm{H}_{33} \mathrm{O}_{4} \mathrm{P}$ requires $M, 476.2116$ ); $v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 1740(\mathrm{C}=\mathrm{O}), 1612(\mathrm{C}=\mathrm{C}), 1437(\mathrm{P}-\mathrm{Ph})$ and $1176(\mathrm{P}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.9-7.1\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 7.08(2 \mathrm{H}, \mathrm{d}$, $J 8.7, \mathrm{Ar}), 6.68(2 \mathrm{H}, \mathrm{d}, J 8.7, \mathrm{Ar}), 6.20\left(1 \mathrm{H}, \mathrm{dd}, J 4.5\right.$ and ${ }^{3} J_{\mathrm{PH}}$ $8.5, \mathrm{CHOAc}), 5.05(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CH}), 3.71(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.86$ $(1 \mathrm{H}, \mathrm{m}), 2.52(2 \mathrm{H}, \mathrm{m}), 1.98(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 1.9-1.75(2 \mathrm{H}, \mathrm{m}), 1.15$ $(2 \mathrm{H}, \mathrm{m})$ and $0.82(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $169.2^{-}$(C=O), $159.0^{-}$(ipso-Ar), 132-127 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar), $113.4^{+}(\mathrm{Ar}), 72.7^{+}$(CHOAc), 55.1+ $(\mathrm{OMe})$, $45.1^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.1, \mathrm{PCH}\right), 34.2^{-}, 28.6^{-}, 27.7^{-}, 22.2^{-}, 20.8^{+}$(Ac) and $13.6^{+}(\mathrm{Me}) ; m / z 476.2\left(15 \%, \mathrm{M}^{+}\right)$, 433 ( $40, \mathrm{M}-\mathrm{Ac}$ ), 277 (100) and 201.1 ( $35, \mathrm{Ph}_{2} \mathrm{PO}$ ).

## ( $\mathbf{2} R^{*}, 3 S^{*}$ )-2-Benzoyloxy-3-diphenylphosphinoylbutanoic acid 26

Sodium periodate ( $1.15 \mathrm{~g}, 5.4 \mathrm{mmol}$ ) was added to a stirred solution of the benzoate $25(181 \mathrm{mg}, 0.39 \mathrm{mmol})$ in $3: 2: 2$ water-acetonitrile-carbon tetrachloride $\left(3.5 \mathrm{~cm}^{3}\right)$. The reaction mixture was stirred until the phases were clear, ruthenium chloride ( $2 \mathrm{mg}, 1 \mu \mathrm{~mol}$ ) added, the reaction mixture stirred for 16 h , quenched with water $\left(10 \mathrm{~cm}^{3}\right)$ extracted with dichloromethane $\left(3 \times 10 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was the acid 26, $R_{\mathrm{f}} 0.0$ (EtOAc) (Found: $\mathrm{MH}^{+}$, 409.1210. $\mathrm{C}_{23} \mathrm{H}_{21} \mathrm{O}_{5} \mathrm{P}$ requires MH, 409.1205); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3600-2300(\mathrm{OH}), 1724$ ( $\mathrm{C}=\mathrm{O}$ ), $1438(\mathrm{P}-\mathrm{Ph})$ and $1199(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $10.25\left(1 \mathrm{H}\right.$, br s, $\left.\mathrm{CO}_{2} \mathrm{H}\right), 7.95-7.25\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$)$, $5.59\left(1 \mathrm{H}, \mathrm{dd}, J 1.9\right.$ and $\left.^{3} J_{\mathrm{PH}} 9.8, \mathrm{CHOBz}\right), 3.39(1 \mathrm{H}, \mathrm{dqd}, J 1.9$, 7.2 and $\left.{ }^{2} J_{\mathrm{PH}} 10.5, \mathrm{PCH}\right)$ and $1.38\left(3 \mathrm{H}, \mathrm{dd}, J 7.2\right.$ and ${ }^{3} J_{\mathrm{PH}} 15.8$, $\mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 170.2^{-}, 169.9^{-}, 165.1^{-}, 133-128(\mathrm{~m}$, $\mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar ), $70.6^{+}(\mathrm{CHOBz}), 35.0^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 69.9\right.$, $\mathrm{PCH})$ and $7.4^{+}(\mathrm{Me}) ; m / z(\mathrm{FAB}) 409.1\left(100 \%, \mathrm{MH}^{+}\right)$.

## ( $1 R^{*}, 2 S^{*}$ )-2-Diphenylphosphinoyl-1-(4-methoxyphenyl)butane-1,4-diyl dibenzoate 15

By the same general method, the diether $\mathbf{1 4}(55 \mathrm{mg}, 95 \mu \mathrm{~mol})$, sodium periodate ( $280 \mathrm{mg}, 1.31 \mathrm{mmol}$ ) and ruthenium trichloride ( $3 \mathrm{mg}, 1.5 \mu \mathrm{~mol}$ ) gave a crude product which was purified by flash chromatography eluting with $3: 1$ EtOAchexane to give the diester $\mathbf{1 5}(8 \mathrm{mg}, 14 \%)$ as an oil, $R_{\mathrm{f}} 0.52$ (EtOAc); $\delta_{\mathrm{H}}\left(200 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.1-7.1\left(22 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and $2 \times \mathrm{Bz}), 6.73(2 \mathrm{H}, \mathrm{d}, J 6.7, \mathrm{Ar}), 6.35\left(1 \mathrm{H}, \mathrm{dd}, J 3.5\right.$ and $^{2} J_{\mathrm{PH}}$ 9.0, ArCH$), 4.02\left(2 \mathrm{H}, \mathrm{t}, J 5.1, \mathrm{CH}_{2} \mathrm{OBz}\right), 3.68(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $3.19(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH})$ and $2.50(2 \mathrm{H}, \mathrm{m})$.

## ( $2 R^{*}, 3 S^{*}$ )-2-Benzoyloxy-3-diphenylphosphinoylheptanoic acid 18

By the same general method, ( $1 R^{*}, 2 S^{*}$ )-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexyl benzoate 17 ( $401 \mathrm{mg}, 0.78$ $\mathrm{mmol})$, sodium periodate $(2.54 \mathrm{~g}, 11.8 \mathrm{mmol})$ and ruthenium trichloride ( $3 \mathrm{mg}, 1.5 \mu \mathrm{~mol}$ ) gave a crude product which was
dissolved in ethyl acetate $\left(10 \mathrm{~cm}^{3}\right)$ and extracted with saturated sodium bicarbonate solution $\left(3 \times 5 \mathrm{~cm}^{3}\right)$. The combined aqueous fractions were acidifed to pH 1 , extracted with dichloromethane $\left(3 \times 5 \mathrm{~cm}^{3}\right)$ and the organic fractions were washed with brine ( $10 \mathrm{~cm}^{3}$ ) to give the acid $\mathbf{1 8}(320 \mathrm{mg}, 91 \%)$ as an oil, $R_{\mathrm{f}} 0.0$ (EtOAc) (Found: $\mathrm{M}^{+}, 450.1624 . \mathrm{C}_{26} \mathrm{H}_{27} \mathrm{O}_{5} \mathrm{P}$ requires $M$, 450.1599); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3600-2800(\mathrm{OH}), 1721(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 10.25\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CO}_{2} \mathrm{H}\right)$, 8.0-7.21 ( $15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), $5.52\left(1 \mathrm{H}\right.$, br d, ${ }^{3} J_{\mathrm{PH}} 9.8$, CHOBz), $3.32(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 2.1-0.8(6 \mathrm{H}, \mathrm{m})$ and $0.66(3 \mathrm{H}$, $\mathrm{t}, J 6.7, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 170.6^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 5.6, \mathrm{CO}_{2} \mathrm{H}\right)$, $164.9^{-}$( PhCO ), $133-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar$), 69.4^{+}(\mathrm{CHOBz})$, $39.7^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 69.0, \mathrm{PCH}\right), 30.1^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 8.1\right), 23.4^{-}, 22.3^{-}$and $13.4^{+}(\mathrm{Me}) ; m / z 450.2\left(1 \%, \mathrm{M}^{+}\right)$, 202.1 ( $100, \mathrm{Ph}_{2} \mathrm{POH}$ ) and 105.0 ( $95, \mathrm{PhCO}$ ).

## Methyl ( $2 R^{*}, 3 S^{*}$ )-2-benzoyloxy-3-diphenylphosphinoylheptanoate 19

By the same general method, ( $1 R^{*}, 2 S^{*}$ )-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexyl benzoate $17(1.00 \mathrm{~g}, 1.96$ $\mathrm{mmol})$, sodium periodate ( $6.34 \mathrm{~g}, 29.6 \mathrm{mmol}$ ) and ruthenium trichloride ( $10 \mathrm{mg}, 5 \mu \mathrm{~mol}$ ) gave a crude product ( 950 mg ). A portion of this product ( 143 mg ) was dissolved in methanol $\left(5 \mathrm{~cm}^{3}\right)$ and thionyl chloride ( $10 \mu \mathrm{l}, 0.14 \mathrm{mmol}$ ) was added. The reaction mixture was stirred for 36 h , quenched with saturated sodium bicarbonate $\left(5 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane $\left(3 \times 5 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a crude product which was purified by flash chromatography eluting with 1:1 EtOAc-hexane, to give the ester $19(75 \mathrm{mg}, 55 \%)$ as an oil, $R_{\mathrm{f}} 0.59$ (EtOAc) (Found: M ${ }^{+}$, 464.1722. $\mathrm{C}_{27} \mathrm{H}_{29} \mathrm{O}_{5} \mathrm{P}$ requires $M, 464.1725) ; v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1720(\mathrm{C}=\mathrm{O})$ and $1437(\mathrm{P}-\mathrm{Ph})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.0-7.2\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph ), 5.51 ( $1 \mathrm{H}, \mathrm{dd}, J 1.8$ and $\left.{ }^{3} J_{\mathrm{PH}} 11.3, \mathrm{C} H \mathrm{OBz}\right), 3.74(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.16$ $\left(1 \mathrm{H}, \mathrm{dtd}, J 1.8,6.6\right.$ and $\left.^{2} J_{\mathrm{PH}} 11.2, \mathrm{PCH}\right), 1.96(2 \mathrm{H}, \mathrm{m}), 1.25$ $(4 \mathrm{H}, \mathrm{m})$ and $0.79(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $169.9^{-}$(d, ${ }^{3} J_{\mathrm{PC}} 15.9, \mathrm{CO}_{2} \mathrm{Me}$ ), $165.3^{-}$( PhCO ), 133-126 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ar ), $69.8^{+}(\mathrm{CHOBz}), 52.7^{+}(\mathrm{OMe}), 40.5^{+}$(d, $\left.{ }^{1} J_{\mathrm{PC}} 68.5, \mathrm{PCH}\right), 30.5^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 8.6\right), 23.7^{-}, 22.5^{-}$and $13.7^{+}$ (Me); $m / z 464.2$ ( $10 \%,^{+}$), 202.1 (70, $\mathrm{Ph}_{2} \mathrm{POH}$ ), 105.0 ( 100, $\mathrm{PhCO})$ and $77(65, \mathrm{Ph})$.

## ( $2 R^{*}, 3 R^{*}$ )-2-Benzoyloxy-3-diphenylphosphinoyl- $N$-methoxy- $N$ methylheptanamide 23

By the same general method, ( $1 R^{*}, 2 R^{*}$ )-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexyl benzoate 21 ( $380 \mathrm{mg}, 0.74$ $\mathrm{mmol})$, sodium periodate $(2.31 \mathrm{~g}, 10.8 \mathrm{mmol})$ and ruthenium trichloride ( $4 \mathrm{mg}, 2 \mu \mathrm{~mol}$ ) gave a crude product ( 400 mg ). A portion of this product ( 180 mg ) was dissolved in dry dichloromethane ( $5 \mathrm{~cm}^{3}$ ), carbonyldiimidazole ( $71 \mathrm{mg}, 0.44 \mathrm{mmol}$ ) was added, the mixture was stirred for 10 min and N -methoxy- N methylammonium chloride ( $46 \mathrm{mg}, 0.46 \mathrm{mmol}$ ) was added. The reaction mixture was stirred for 16 h , diluted with dichloromethane ( $10 \mathrm{~cm}^{3}$ ), washed with dilute hydrochloric acid ( 0.3 $\mathrm{mol} \mathrm{dm}{ }^{-3}, 2 \times 10 \mathrm{~cm}^{3}$ ), saturated sodium bicarbonate solution $\left(10 \mathrm{~cm}^{3}\right)$ and brine $\left(10 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a crude product which was purified by flash chromatography eluting with 3:1 EtOAc-hexane to give the amide 23 ( $33 \mathrm{mg}, 18 \%$ ) as an oil, $R_{\mathrm{f}} 0.32$ (EtOAc) (Found: $\mathrm{M}^{+}-$ MeONMe , 433.1574. $\mathrm{C}_{28} \mathrm{H}_{32} \mathrm{NO}_{5} \mathrm{P}$ requires $\mathrm{M}-\mathrm{MeONMe}$, 433.1568); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1718(\mathrm{C}=\mathrm{O}), 1664$ (amide $\mathrm{C}=\mathrm{O}$ ), $1422(\mathrm{P}-\mathrm{Ph})$ and $1213(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.03-7.79$ $(4 \mathrm{H}, \mathrm{m}), 7.48-7.1\left(11 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and remaining Ph$), 6.02$ $(1 \mathrm{H}, \mathrm{dd}, J 8.4$ and $12.8, \mathrm{CHOBz}), 3.86(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.21$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 3.13(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 1.69(2 \mathrm{H}, \mathrm{m}), 1.4-1.0(4 \mathrm{H}$ $\mathrm{m})$ and $0.71(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 168.7^{-}$, $165.8^{-}, 134-127\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar), $69.4^{+}$(CHOBz), 61.1 ${ }^{+}$ (OMe), $39.8^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 69.4, \mathrm{PCH}\right), 30.2^{+}(\mathrm{NMe}), 30.4^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}}\right.$ 7.5), $24.4^{-}, 22.5^{-}$and $13.5^{+}(\mathrm{Me}) ; \mathrm{m} / \mathrm{z} 433.2\left(20 \%, \mathrm{M}^{+}-\right.$ $\mathrm{MeONMe})$ and $105.0(100, \mathrm{PhCO})$.

## Methyl ( $2 R^{*}, 3 S^{*}$ )-3-diphenylphosphinoyl-2-hydroxyheptanoate

 29By the same general method, $\left(1 R^{*}, 2 S^{*}\right)$-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexyl acetate 27 ( $505 \mathrm{mg}, 1.12$ $\mathrm{mmol})$, sodium periodate $(4.0 \mathrm{~g}, 18.7 \mathrm{mmol})$ and ruthenium trichloride ( $40 \mathrm{mg}, 0.2 \mathrm{mmol}$ ) gave a crude product ( 550 mg ) which was the acid 28, $R_{\mathrm{f}} 0.0$ (EtOAc) (Found: $\mathrm{M}^{+}, 390.1626$. $\mathrm{C}_{26} \mathrm{H}_{29} \mathrm{O}_{4} \mathrm{P}$ requires $\left.M, 390.1596\right)$; $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3700-$ $2700(\mathrm{OH}), 1744(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1205(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.9-7.3\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 6.4\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CO}_{2} \mathrm{H}\right)$, $5.52\left(1 \mathrm{H}\right.$, dd, $J 2.5$ and $\left.{ }^{3} J_{\mathrm{PH}} 14.9, \mathrm{C} H \mathrm{OAc}\right), 3.19(1 \mathrm{H}, \mathrm{m}$, $\mathrm{PCH}), 2.0-1.1(6 \mathrm{H}, \mathrm{m}), 1.79(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ and $0.75(3 \mathrm{H}, \mathrm{t}$, $J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 170.4^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 13.4, \mathrm{C}=\mathrm{O}\right)$, $169.6^{-}(\mathrm{MeCO}), 132-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 69.5^{+}$(CHOAc), $39.8^{+}$ (d, ${ }^{1} J_{\mathrm{PC}} 69.0, \mathrm{PCH}$ ), $30.3^{-}$(d, ${ }^{2} J_{\mathrm{PC}} 10.0$ ), $23.9^{-}, 22.4^{-}, 20.2^{-}$ (OAc) and $13.6^{+}(\mathrm{Me}) ; m / z 390.2\left(10 \%, \mathrm{M}^{+}\right), 294(100)$ and 69.3 (100). A portion of the crude product ( 148 mg ) was dissolved in methanol ( $10 \mathrm{~cm}^{3}$ ), concentrated hydrochloric acid was added $\left(4 \mathrm{~cm}^{3}\right)$ and the reaction stirred at $50{ }^{\circ} \mathrm{C}$ for 1 day. The reaction was quenched with water $\left(10 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a crude product, which was purified by flash chromatography eluting with 5\% methanol in EtOAc to give the ester 29 ( 68 mg , $59 \%$ ) as an oil, $R_{\mathrm{f}} 0.40$ (EtOAc) (Found: $\mathrm{M}^{+}, 360.1490$. $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{O}_{4} \mathrm{P}$ requires $M, 360.1490$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3387$ $(\mathrm{OH}), 1732(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1221(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $7.95-7.4$ ( $10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ ), $4.63\left(1 \mathrm{H}, \mathrm{d},{ }^{3} J_{\mathrm{PH}} 12.1\right.$, $\mathrm{CHOH}), 4.31(1 \mathrm{H}, \mathrm{brs}, \mathrm{OH}), 3.73(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.88(1 \mathrm{H}, \mathrm{m}$, PCH), $1.92(1 \mathrm{H}, \mathrm{m}), 1.55(1 \mathrm{H}, \mathrm{m}), 1.3-0.9(4 \mathrm{H}, \mathrm{m})$ and 0.71 $(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 173.1^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 17.4\right.$, $\mathrm{C}=\mathrm{O}$ ), $132-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 68.8^{+}, 52.3^{+}(\mathrm{OMe}), 41.0^{+}(\mathrm{d}$, ${ }^{1} J_{\mathrm{PC}} 68.9, \mathrm{PCH}$ ), $30.4^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 9.5\right.$ ), $22.5^{-}, 22.4^{-}$and $13.6^{+}$ $(\mathrm{Me}) ; m / z 360.1\left(10 \%, \mathrm{M}^{+}\right), 245.1$ (60) and 202.1 (100, $\left.\mathrm{Ph}_{2} \mathrm{POH}\right)$.

## Methyl ( $2 R^{*}, 3 S^{*}$ )-3-diphenylphosphinoyl-2-hydroxyheptanoate 32

By the same general method, $\left(1 R^{*}, 2 R^{*}\right)$-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)hexyl acetate 30 ( $1.02 \mathrm{~g}, 2.26$ $\mathrm{mmol})$, sodium periodate $(8.0 \mathrm{~g}, 37.0 \mathrm{mmol})$ and ruthenium trichloride ( $100 \mathrm{mg}, 0.5 \mathrm{mmol}$ ) gave a crude product $(0.91 \mathrm{~g})$ which was the acid 31, $R_{\mathrm{f}} 0.0$ (EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{H}$, 389.1511. $\mathrm{C}_{26} \mathrm{H}_{29} \mathrm{O}_{4} \mathrm{P}$ requires $\left.M-H, 389.1514\right) ; v_{\text {max }} / \mathrm{cm}^{-1}$ $\left(\mathrm{CHCl}_{3}\right) 3500-2500(\mathrm{OH}), 1742(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and 1195 ( $\mathrm{P}=\mathrm{O}$ ); $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 10.5-10.0\left(1 \mathrm{H}\right.$, br s, $\left.\mathrm{CO}_{2} \mathrm{H}\right)$, 7.9-7.4 $\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 5.53\left(1 \mathrm{H}, \mathrm{dd}, J 2.3\right.$ and ${ }^{3} J_{\mathrm{PH}} 17.3$, CHOAc), 3.14 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}$ ), 1.81 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OAc}$ ), $1.65-1.1$ $(6 \mathrm{H}, \mathrm{m})$ and $0.73(3 \mathrm{H}, \mathrm{t}, J 7.1, \mathrm{Me})$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $169.5^{-}, 168.9^{-}, 133-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 70.1^{+}$(CHOAc), $40.5^{+}$ (d, ${ }^{1} J_{\mathrm{PC}} 67.1, \mathrm{PCH}$ ), $30.0^{-}$(d, ${ }^{2} J_{\mathrm{PC}} 10.9$ ), 25.1 $1^{-}, 21.9^{-}, 20.1^{+}$ (OAc) and $13.4^{+}(\mathrm{Me}) ; m / z 389.2\left(10 \%, \mathrm{M}^{+}-\mathrm{H}\right)$ and 201.0 $\left(\mathrm{Ph}_{2} \mathrm{PO}\right)$. The crude product was dissolved in methanol $\left(60 \mathrm{~cm}^{3}\right)$, concentrated hydrochloric acid was added $\left(20 \mathrm{~cm}^{3}\right)$ and the reaction was stirred at $50^{\circ} \mathrm{C}$ for 1 day. The reaction was quenched with water $\left(60 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane $\left(3 \times 50 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a crude product, which was purified by flash chromatography eluting with EtOAc to give the ester 32 ( $432 \mathrm{mg}, 55 \%$ ) as an oil, $R_{\mathrm{f}}$ 0.38 (EtOAc) (Found: $\mathrm{M}^{+}, 360.1490 . \mathrm{C}_{20} \mathrm{H}_{25} \mathrm{O}_{4} \mathrm{P}$ requires $M$, 360.1490); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3370(\mathrm{OH}), 1733(\mathrm{C}=\mathrm{O})$ and $1438(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.9-7.3(10 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{Ph}_{2} \mathrm{PO}\right), 5.05(1 \mathrm{H}, \mathrm{d}, J 8.8, \mathrm{OH}), 4.50(1 \mathrm{H}$, ddd, $J 2.8,8.7$ and $\left.{ }^{3} J_{\mathrm{PH}} 26.8, \mathrm{CHOH}\right), 3.15(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 2.84(1 \mathrm{H}$, tdd, $J 2.7,8.0$ and $1.08, \mathrm{PCH}), 2.0-1.1(6 \mathrm{H}, \mathrm{m})$ and $0.75(3 \mathrm{H}$, $\mathrm{t}, J 6.9, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 173.2^{-}, 132-128(\mathrm{~m}$, $\mathrm{Ph}_{2} \mathrm{PO}$ ), $70.5^{+}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 4.5, \mathrm{CHOH}\right), 51.6^{+}(\mathrm{OMe}), 40.0^{+}(\mathrm{d}$, $\left.{ }^{1} J_{\mathrm{PC}} 66.8, \mathrm{PCH}\right), 29.5^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 11.1\right), 25.5^{-}, 22.0^{-}$and $13.0^{+}$ (Me); $m / z 360.1\left(50 \%, \mathrm{M}^{+}\right)$, $301.1\left(65, \mathrm{M}^{+}-\mathrm{MeCO}\right)$ and 201.1 ( $100, \mathrm{Ph}_{2} \mathrm{PO}$ ).

## Methyl ( $2 R^{*}, 3 S^{*}$ )-3-diphenylphosphinoyl-2-hydroxy-5phthalimidopentanoate 38

By the same general method, $\left(1 R^{*}, 2 S^{*}\right)$-2-diphenylphos-phinoyl-1-(4-methoxyphenyl)-4-phthalimidobutyl acetate 36 ( $978 \mathrm{mg}, 1.72 \mathrm{mmol}$ ), sodium periodate ( $7.62 \mathrm{~g}, 35.6 \mathrm{mmol}$ ) and ruthenium trichloride ( $10 \mathrm{mg}, 5 \mu \mathrm{~mol}$ ) gave a crude product which was the acid 37, $R_{\mathrm{f}} 0.0(\mathrm{EtOAc})$ (Found: $\mathrm{MH}^{+}, 506.1369$. $\mathrm{C}_{27} \mathrm{H}_{24} \mathrm{NO}_{7} \mathrm{P}$ requires $\left.M H, 506.1398\right)$; $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3800-$ $2500(\mathrm{OH}), 1740-1650(\mathrm{C}=\mathrm{O})$ and $1206(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.70\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CO}_{2} \mathrm{H}\right), 7.9-7.4\left(14 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar), $5.45\left(1 \mathrm{H}, \mathrm{dd}, J 1.4\right.$ and $\left.^{3} J_{\mathrm{PH}} 13.2, \mathrm{CHOAc}\right), 3.58(1 \mathrm{H}, \mathrm{m})$, $3.35(1 \mathrm{H}, \mathrm{m}), 2.5-2.0(3 \mathrm{H}, \mathrm{m})$ and $1.98(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$; $\delta_{\mathrm{C}}(100$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $170.4^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 13.2, \mathrm{C}=\mathrm{O}\right), 169.0^{-}, 168.0^{-}, 133.8^{+}$ (Ar), 133-128 ( $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar), $128.2^{+}$(Ar), $116.4^{-}(\mathrm{Ar}), 68.7^{+}$(CHOAc), $37.9^{+}$(d, ${ }^{1} J_{\mathrm{PC}} 70.8, \mathrm{PCH}$ ), $36.8^{-}$ (d, ${ }^{2} J_{\mathrm{PC}} 9.7$ ), $23.4^{-}$and $20.3^{+}$(OAc); $m / z$ (FAB) 506.1 ( $80 \%$, $\mathrm{MH}^{+}$) and 307.1 (100). The crude product was dissolved in methanol ( $60 \mathrm{~cm}^{3}$ ), concentrated hydrochloric acid was added $\left(20 \mathrm{~cm}^{3}\right)$ and the reaction was stirred at $50{ }^{\circ} \mathrm{C}$ for 5 h . The reaction was quenched with water $\left(60 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane ( $3 \times 50 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated to give a crude product, which was purified by flash chromatography eluting with EtOAc to give the ester $38(412 \mathrm{mg}$, $61 \%$ ) as an oil, $R_{\mathrm{f}} 0.38$ (EtOAc) (Found: $\mathrm{MH}^{+}$, 478.1538. $\mathrm{C}_{26} \mathrm{H}_{24} \mathrm{NO}_{6} \mathrm{P}$ requires MH, 478.1419); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3449$ $(\mathrm{OH}), 1710(\mathrm{C}=\mathrm{O}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1219(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 8.0-7.3\left(14 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar$), 4.66(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $4.62\left(1 \mathrm{H}, \mathrm{d},{ }^{3} J_{\mathrm{PH}} 10.5, \mathrm{CHOH}\right), 3.60(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.41(1 \mathrm{H}$, td, $J 5.5$ and $\left.14.0, \mathrm{NCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.22(1 \mathrm{H}$, ddd, $J 5.7,8.9$ and $\left.14.0, \mathrm{NCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 3.01(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH})$ and $2.2(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(100$ $\left.\mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ 168.5-( $\mathrm{C}=\mathrm{O} \times 3$ ), 134-128 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar), $123.1^{+}(\mathrm{Ar}), 113.6^{+}(\mathrm{Ar}), 70.2^{+}, 52.6^{+}(\mathrm{OMe})$, $38.2^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 68.7, \mathrm{PCH}\right), 36.9^{-}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 5.3\right)$ and $22.5^{-}$; $m / z 478.1\left(\mathrm{FAB}, \mathrm{MH}^{+}\right), 279$ (100, EI) and 201.1 (90, EI, $\mathrm{Ph}_{2} \mathrm{PO}$ ).

## Methyl ( $2 R^{*}, 3 S^{*}$ )-2-benzoyloxy-3-diphenylphosphinoylheptanoate 19

Diazomethane ${ }^{30}$ (ca. $0.3 \mathrm{~mol} \mathrm{dm}^{-3}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) was added dropwise to a solution of acid $\mathbf{1 8}$ in ethyl acetate ( $6 \mathrm{~cm}^{3}$ ) until a yellow colour persisted. The reaction was quenched by dropwise addition of acetic acid until the reaction mixture was colourless, and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with EtOAc, to give the methyl ester 19 ( $150 \mathrm{mg}, 44 \%$ ), identical spectroscopically with that obtained previously.

## ( $2 R^{*}, 3 S^{*}$ )-3-Diphenylphosphinoylheptane-1,2-diol 33

Methyl ester 19 ( $260 \mathrm{mg}, 056 \mathrm{mmol}$ ) in dry THF ( $5 \mathrm{~cm}^{3}$ ) was added by cannula dropwise to a stirred suspension of lithium aluminium hydride ( $127 \mathrm{mg}, 3.36 \mathrm{mmol}$ ) in dry THF $\left(5 \mathrm{~cm}^{3}\right)$ at $20^{\circ} \mathrm{C}$. The reaction was stirred for 2 h , quenched with water $\left(10 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $5 \%$ methanol in EtOAc to give the diol 33 ( 96 mg , $52 \%$ ) as an oil, $R_{\mathrm{f}} 0.23$ ( $5 \%$ methanol in EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{CH}_{2} \mathrm{OH}, 301.1354 . \mathrm{C}_{19} \mathrm{H}_{25} \mathrm{O}_{3} \mathrm{P}$ requires $\mathrm{M}-\mathrm{CH}_{2} \mathrm{OH}$, 301.1357); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3387(\mathrm{OH}), 1438(\mathrm{P}-\mathrm{Ph})$ and 1223 $(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.9-7.7(4 \mathrm{H}, \mathrm{m}), 7.55-7.4(6 \mathrm{H}$, $\mathrm{m}), 5.17(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOH}), 3.79(1 \mathrm{H}, \mathrm{dd}, J 5.6$ and 11.3 , $\left.\mathrm{C} H_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OH}\right), 3.48\left(1 \mathrm{H}, \mathrm{dd}, J 5.2\right.$ and 11.3, $\left.\mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OH}\right), 3.38$ $(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 2.47(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 1.84(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $1.8-1.0(6 \mathrm{H}, \mathrm{m})$ and $0.72\left(3 \mathrm{H}, \mathrm{t}, J 7.3\right.$, Me); $\delta_{\mathrm{C}}(100 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) 133-128 (m, Ph ${ }_{2} \mathrm{PO}$ ), 70.8 ${ }^{+}(\mathrm{CHOH}), 63.5^{-}\left(\mathrm{d},{ }^{3} \mathrm{JPC}_{\mathrm{PC}}\right.$ 12.1, $\mathrm{CH}_{2} \mathrm{OH}$ ), $40.2^{+}$(d, ${ }^{1} J_{\mathrm{PC}} 68.9, \mathrm{PCH}$ ), $32.1^{-}, 32.9^{-}, 22.5^{-}$ and $13.5^{+}(\mathrm{Me})$; $m / z 301.1\left(20 \%, \mathrm{M}^{+}-\mathrm{CH}_{2} \mathrm{OH}\right)$, 201.1 (100, $\left.\mathrm{Ph}_{2} \mathrm{PO}\right)$ and $77.0(60, \mathrm{Ph})$.

Methyl ( $2 R^{*}, 3 R^{*}$ )-3-diphenylphosphinoyl-2-methoxymethoxyheptanoate 39
Methoxymethyl chloride ( $0.22 \mathrm{~cm}^{3}, 0.29 \mathrm{mmol}$ ) and diisopropylethylamine ( $0.42 \mathrm{~cm}^{3}, 2.4 \mathrm{mmol}$ ) were added dropwise to a stirred solution of methyl $\left(2 R^{*}, 3 S^{*}\right)$-3-diphenylphosphinoyl-2hydroxyheptanoate 32 ( $218 \mathrm{mg}, 0.61 \mathrm{mmol}$ ) in dry dichloromethane $\left(5 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$. The solution was stirred for 1 h , allowed to warm to room temperature, stirred for a further 7 days, quenched with aqueous saturated sodium carbonate $\left(10 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane ( $3 \times 5 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude which was purified by flash chromatography eluting with EtOAc to give the methoxymethyl ether $39(210 \mathrm{mg}, 86 \%)$ as an oil, $R_{\mathrm{f}} 0.38$ (EtOAc) (Found: $\mathrm{M}^{+}, 404.1742 . \mathrm{C}_{22} \mathrm{H}_{29} \mathrm{O}_{5} \mathrm{P}$ requires $M, 404.1752$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1743(\mathrm{C}=\mathrm{O}), 1437(\mathrm{P}-\mathrm{Ph})$, $1182(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.8-7.2\left(10 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right)$, $4.46\left(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{MeOCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.42\left(1 \mathrm{H}\right.$, dd, $J 6.6$ and $^{3} J_{\mathrm{PH}}$ 13.2, CHOMOM), $4.34\left(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{MeOCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 3.59(3 \mathrm{H}$, s, OMe), $2.99(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}), 1.7-1.0(6 \mathrm{H}, \mathrm{m})$ and $0.70(3 \mathrm{H}, \mathrm{t}$, $J 7.2, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 133-128\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right), 96.7^{-}$ $\left(\mathrm{OCH}_{2} \mathrm{O}\right), 75.3^{+}(\mathrm{CHOMOM}), 56.2^{+}, 52.0^{+}(\mathrm{OMe} \times 2), 41.5^{+}$ (d, ${ }^{1} J_{\mathrm{PC}} 69.8, \mathrm{PCH}$ ), $29.8^{-}$(d, ${ }^{2} J_{\mathrm{PC}} 8.2$ ), 25.3-, $22.4^{-}$and $13.6^{+}$ (Me); $m / z 404.2\left(20 \%, \mathrm{M}^{+}\right), 202.1\left(100, \mathrm{Ph}_{2} \mathrm{POH}\right)$ and 201.1 (100, $\mathrm{Ph}_{2} \mathrm{PO}$ ).

## Methyl ( $2 R^{*}, 3 S^{*}$ )-3-diphenylphosphinoyl-2-methoxymethoxy-5phthalimidopentanoate 40

By the same general method, methyl $\left(2 R^{*}, 3 S^{*}\right)$-3-diphenyl-phosphinoyl-2-hydroxy-5-phthalimidopentanoate 38 ( 375 mg , 0.89 mmol ), diisopropylethylamine ( $1.26 \mathrm{~cm}^{3}, 7.2 \mathrm{mmol}$ ) and methoxymethyl chloride ( $0.66 \mathrm{~cm}^{3}, 0.89 \mathrm{mmol}$ ) gave a crude product after 10 days which was purified by flash chromatography eluting with $2 \%$ methanol in EtOAc to give the methoxymethyl ether 40 ( $362 \mathrm{mg}, 87 \%$ ) as an oil, $R_{\mathrm{f}} 0.29$ (EtOAc) (Found: $\mathrm{MH}^{+}$, 522.1661. $\mathrm{C}_{28} \mathrm{H}_{28} \mathrm{NO}_{7} \mathrm{P}$ requires MH , 522.1682 ); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 1772$ (imide $\left.\mathrm{C}=\mathrm{O}\right), 1747(\mathrm{C}=\mathrm{O})$, 1712 (imide $\mathrm{C}=\mathrm{O}$ ), 1438 ( $\mathrm{P}-\mathrm{Ph}$ ) and 1173 ( $\mathrm{P}=\mathrm{O}$ ); $\delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\mathrm{CDCl}_{3}$ ) $7.85-7.2\left(14 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ar$), 4.62(3 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CHOCH} \mathrm{O}_{2} \mathrm{OMe}\right)$, $3.63(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe})$, $3.45(2 \mathrm{H}, \mathrm{m}), 3.26$ ( 3 $\mathrm{H}, \mathrm{m}, \mathrm{OMe}), 2.94(1 \mathrm{H}, \mathrm{m})$ and $2.45-2.1(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}(50$ $\mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $171.9^{-}\left(\mathrm{d},{ }^{3} \mathrm{~J}_{\mathrm{PC}} 14.4, \mathrm{C}=\mathrm{O}\right), 167.9^{-}(\mathrm{C}=\mathrm{O})$, 134-128 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and remaining Ar), $122.9^{+}$(Ar), 96.4 $\left(\mathrm{OCH}_{2} \mathrm{O}\right), 72.1^{+}(\mathrm{CHOMOM}), 56.3^{+}, 52.1^{+}(\mathrm{OMe} \times 2), 39.5^{+}$ (d, $\left.{ }^{1} J_{\mathrm{PC}} 69.2, \mathrm{PCH}\right), 37.0^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 9.3\right)$ and $22.3^{-} ; \mathrm{m} / \mathrm{z}$ (FAB) $522.1\left(100 \%, \mathrm{MH}^{+}\right), 279(80), 201.1\left(75, \mathrm{Ph}_{2} \mathrm{PO}\right)$ and 154.1 (100).

## $\left(2 R^{*}, 6 R^{*}, 1^{\prime} S^{*}\right)$ - and $\left(2 S^{*}, 6 R^{*}, 1^{\prime} R^{*}\right)$-2-(3'-Benzyloxy-1'-di-phenylphosphinoylpropyl)-6-hydroxy-3,6-dihydro-2H-pyran-3one 47

$m$-Chloroperbenzoic acid ( $328 \mathrm{mg}, 57-85 \%$ by weight, $c a .1 .34$ $\mathrm{mmol})$ was added to a stirred solution of $\left(1 R^{*}, 2 S^{*}\right)-4$-benzyl-oxy-2-diphenylphosphinoyl-1-(2-furyl)butan-1-ol ${ }^{1} 46$ ( 596 mg , $1.34 \mathrm{mmol})$ in dry dichloromethane ( $12 \mathrm{~cm}^{3}$ ) at $0{ }^{\circ} \mathrm{C}$. The reaction mixture was stirred for 2 days, diluted with dichloromethane ( $10 \mathrm{~cm}^{3}$ ) washed with saturated sodium carbonate solution ( $20 \mathrm{~cm}^{3}$ ), saturated sodium thiosulfate solution ( $20 \mathrm{~cm}^{3}$ ) and saturated sodium carbonate solution ( $20 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with 5\% methanol in EtOAc to give the enone 47 (401 $\mathrm{mg}, 65 \%, 83: 17$ mixture of hemiacetal epimers) as a foam, $R_{\mathrm{f}}$ 0.25 (EtOAc) (Found: $\mathrm{MH}^{+}$, 463.1644. $\mathrm{C}_{27} \mathrm{H}_{27} \mathrm{O}_{5} \mathrm{P}$ requires MH, 463.1674); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3268(\mathrm{br}, \mathrm{OH}), 1696(\mathrm{C}=\mathrm{O})$, $1628(\mathrm{C}=\mathrm{C}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1166(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right)$ 8.0-7.8 ( $4 \mathrm{H}, \mathrm{m}$ ), 7.6-7.2 (11 H, m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), $6.96\left(1 \mathrm{H}, \mathrm{br} \mathrm{d}, J 10.2, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {min }}\right), 6.85(1 \mathrm{H}, \mathrm{dd}, J 3.6$ and $\left.10.2, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {maj }}\right), 6.09\left(1 \mathrm{H}\right.$, br d, $\left.J 10.2, \mathrm{CH}_{\mathrm{A}}=\mathrm{C}_{\mathrm{B}}{ }^{\text {min }}\right)$,
$5.99\left(1 \mathrm{H}, \mathrm{dd}, J 1.0\right.$ and $\left.10.3, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {maj }}\right), 5.62(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\left.\mathrm{OC} H \mathrm{OH}^{\text {maj }+ \text { min }}\right)$, $4.91\left(1 \mathrm{H}, \mathrm{d}, J 13.4, \mathrm{PCHCHO}^{\text {maj }+ \text { min }}\right)$, $4.28\left(1 \mathrm{H}, \mathrm{d}, J 11.9, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}{ }^{\text {maj }}\right), 4.22(1 \mathrm{H}, \mathrm{d}, J 11.9$, $\left.\mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}{ }^{\text {maj }}\right), 3.75\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}} \mathrm{OBn}^{\text {min }}\right), 3.63(1 \mathrm{H}, \mathrm{br} \mathrm{q}$, $J$ 8.1, $\left.\mathrm{CH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}} \mathrm{OBn}^{\text {maj }}\right), 3.4-3.2\left(1 \mathrm{H}, \mathrm{m}, \mathrm{PCH}\right.$ and $\mathrm{CH}_{\mathrm{A}} H_{\mathrm{B}^{-}}$ $\mathrm{OBn}^{\text {maj }+ \text { min }}$ ) and 2.3-2.0 $(2 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $195.7^{-}(\mathrm{C}=\mathrm{O}), 151.0^{+}\left(\mathrm{CH}=\mathrm{CH}^{\text {min }}\right), 146.6^{+}\left(\mathrm{CH}=\mathrm{CH}^{\text {maj }}\right), 138.4^{-}$ (ipso- $\mathrm{Ph}^{\text {maj }+ \text { min }}$ ), 132-126 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), $91.3^{+}$(OCHO$\left.\mathrm{H}^{\text {min }}\right), \quad 87.3^{+}\left(\mathrm{OCHOH}^{\text {maj }}\right), \quad 76.3^{+}\left(\mathrm{PCHCHO}{ }^{\text {min }}\right), 72.5^{-}$ $\left(\mathrm{CH}_{2} \mathrm{Ph}^{\text {maj }+ \text { min }}\right), 68.1^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 11.3, C \mathrm{H}_{2} \mathrm{OBn}^{\text {maj }+ \text { min }}\right), 34.3^{+}(\mathrm{d}$, ${ }^{1} J_{\mathrm{PC}} 72.8, \mathrm{PCH}^{\mathrm{maj}}$ ), $24.1^{-}(\mathrm{min})$ and $23.8^{-}(\mathrm{maj}) ; m / z(\mathrm{FAB})$ $463.3\left(60 \%, \mathrm{MH}^{+}\right)$and 307.1 (100).

## ( $2 S, 6 R, 1^{\prime} R, 3^{\prime} R$ )- and ( $2 S, 6 S, 1^{\prime} R, 3^{\prime} R$ )-2-( $3^{\prime}$-Benzyloxy-1'-di-

 phenylphosphinoylheptyl)-6-hydroxy-3,6-dihydro-2H-pyran-3one 67By the same general method, ( $1 S, 2 R, 4 R$ )-4-benzyloxy-2-diphenylphosphinoyl-1-(2-furyl)octan-1-ol ${ }^{1} 66$ ( $1.52 \mathrm{~g}, 3.02$ mmol ) and $m$-chloroperbenzoic acid ( $840 \mathrm{mg}, 57-85 \%$ by weight, ca. 3.41 mmol ) gave a crude product after 16 h which was purified by flash chromatography eluting with $5 \%$ methanol in EtOAc) to give the enone $\mathbf{6 7}(1.49 \mathrm{~g}, 95 \%, 77: 23$ mixture of hemiacetal epimers) as a foam, $R_{\mathrm{f}} 0.53$ ( $10 \%$ methanol in EtOAc); $[a]_{\mathrm{D}}^{20}+47.5\left(c 0.20\right.$ in $\mathrm{CHCl}_{3}$ ) (Found: $\mathrm{MH}^{+}, 519.2337$. $\mathrm{C}_{31} \mathrm{H}_{35} \mathrm{O}_{5} \mathrm{P}$ requires $\left.\mathrm{MH}, 519.2300\right)$; $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3459$ $(\mathrm{OH}), 1695(\mathrm{C}=\mathrm{O}), 1601(\mathrm{C}=\mathrm{C}), 1423(\mathrm{P}-\mathrm{Ph})$ and $1202(\mathrm{P}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.1-7.2\left(4 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 6.95$ $\left(1 \mathrm{H}, \mathrm{brd}, J 10.3, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {min }}\right), 6.80(1 \mathrm{H}$, dd, $J 3.3$ and 10.2 , $\left.\mathrm{C} H_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {maj }}\right), 6.03\left(1 \mathrm{H}\right.$, br d, $\left.J 10.3, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {min }}\right), 5.98(1 \mathrm{H}$, d, $\left.J 10.2, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}{ }^{\text {maj }}\right), 5.80\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OCHOH}^{\text {maj }}\right), 5.20$ $\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OCHOH}{ }^{\text {min }}\right), 4.98\left(1 \mathrm{H}, \mathrm{d}, J 14.4, \mathrm{PCHCHO}^{\text {maj }+ \text { min }}\right)$, 4.5-4.0 ( $2 \mathrm{H}, \mathrm{m}$ ), $3.60\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOBn}{ }^{\text {maj }+ \text { min }}\right), 3.00(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{PCH}^{\text {maj }+ \text { min }}\right), 2.3(2 \mathrm{H}, \mathrm{m}), 1.9(2 \mathrm{H}, \mathrm{m}), 1.5-1.1(4 \mathrm{H}, \mathrm{m})$ and $0.90\left(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me}^{\mathrm{maj}+\mathrm{min}}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 195.4^{-}$ $\left(\mathrm{C}=\mathrm{O}^{\text {maj }}\right), 150.9^{+}\left(\mathrm{CH}=\mathrm{CH}^{\mathrm{min}}\right), 146.4^{+}\left(\mathrm{CH}=\mathrm{CH}^{\text {maj }}\right), 139.2^{-}$ (ipso- $\left.\mathrm{Ph}^{\text {maj }}\right), 133-126\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}, \mathrm{CH}=\mathrm{CH}\right.$ and Ph$), 91.7^{+}$ $\left(\mathrm{OCHOH}^{\text {min }}\right), 87.5^{+}\left(\mathrm{OCHOH}^{\text {maj }}\right), \quad 76.6^{+}\left(\mathrm{d}^{3} J_{\mathrm{PC}} 10.3\right.$, $\mathrm{CHOBn}), 71.8^{+}, 69.5^{-}\left(\mathrm{CH}_{2} \mathrm{Ph}^{\text {maj }}\right), 60.4^{-}\left(\mathrm{CH}_{2} \mathrm{Ph}^{\text {min }}\right), 36.5^{+}$ (d, $\left.{ }^{1} J_{\mathrm{PC}} 72.4, \mathrm{PCH}^{\text {maj }+ \text { min }}\right), 32.5^{-}, 27.9^{-}, 26.9^{-}(\mathrm{maj}), 22.6^{-}$(d, $\left.J_{\mathrm{PC}} 4.3\right)$ and $14.2^{+}(\mathrm{Me}) ; m / z(\mathrm{FAB}) 519.2\left(65 \%, \mathrm{MH}^{+}\right)$and $201.0\left(100, \mathrm{Ph}_{2} \mathrm{PO}\right)$.

## $\left(4 R^{*}, 5 S^{*}, 6 R^{*}\right)$-8-Benzyloxy-6-diphenylphosphinoyloctane-1,4,5-triol 49

Sodium borohydride ( $563 \mathrm{mg}, 14.8 \mathrm{mmol}$ ) was added to a solution of 47 ( $943 \mathrm{mg}, 2.11 \mathrm{mmol}$ ) in absolute ethanol $\left(5 \mathrm{~cm}^{3}\right)$ and stirred for 16 h at room temperature. Hydrochloric acid ( $3.0 \mathrm{~mol} \mathrm{dm}^{-3}$, five drops) and water $\left(10 \mathrm{~cm}^{3}\right)$ were added to the reaction mixture, the majority of the ethanol removed under reduced pressure, the aqueous suspension extracted with dichloromethane ( $3 \times 10 \mathrm{~cm}^{3}$ ), and the combined organic fractions dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $10 \%$ methanol in EtOAc to give the triol $49(620 \mathrm{mg}, 66 \%)$ as an oil, $R_{\mathrm{f}} 0.21(7 \%$ methanolEtOAc) (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}, 379.1463 . \mathrm{C}_{27} \mathrm{H}_{33} \mathrm{O}_{5} \mathrm{P}$ requires $\left.M-C_{4} H_{9} O_{2}, 379.1463\right) ; v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3425(\mathrm{OH}), 1438$ ( $\mathrm{P}-\mathrm{Ph}$ ) and $1199(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 8.0-7.2(15 \mathrm{H}$, $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and Ph$), 4.57\left(1 \mathrm{H}, \mathrm{d}, J 12.1, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.28(1 \mathrm{H}$, d, $\left.J 12.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 3.79\left(1 \mathrm{H}, \mathrm{t}, J 9.2, \mathrm{OCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.62(1 \mathrm{H}$, quin, $J 5.8), 3.57(2 \mathrm{H}, \mathrm{m}), 3.37(1 \mathrm{H}, \mathrm{qd}, J 1.4$ and $7.0, \mathrm{CHOH})$, $2.84(2 \mathrm{H}, \mathrm{m}, \mathrm{OCH}$ and PCH $), 2.30(1 \mathrm{H}, \mathrm{m}), 1.97(1 \mathrm{H}, \mathrm{m}), 1.83$ $(1 \mathrm{H}, \mathrm{m}), 1.77(2 \mathrm{H}$, quin, $J 6.5)$ and $1.33(1 \mathrm{H}$, quin, $J 8.6)$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 137.3^{-}$(ipso-Ph), $132-127.5\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 73.4^{+}(\mathrm{CHOH}), 72.8^{-}, 70.8^{+}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 11.5, \mathrm{CHOH}\right)$, $69.6^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 4.8, \mathrm{CHOBn}\right), 62.4^{-}, 33.7^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 70.3, \mathrm{PCH}\right)$, $30.2^{-}, 29.1^{-}$and $21.0^{-} ; m / z($ FAB $) 379.2\left(65 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}\right)$, $201.0\left(100, \mathrm{Ph}_{2} \mathrm{PO}\right)$ and $91.0(90, \mathrm{Bn})$.
(4S,5S,6R,8R)-8-Benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol 68
By the same general method, enone $67(1.35 \mathrm{~g}, 2.60 \mathrm{mmol})$ and sodium borohydride ( $691 \mathrm{mg}, 18.2 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with $10 \%$ methanol in EtOAc, to give the triol $68(927 \mathrm{mg}, 68 \%)$ as an oil, $[a]_{\mathrm{D}}^{20}+17.3\left(c 0.21\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}, 435.2102$. $\mathrm{C}_{31} \mathrm{H}_{41} \mathrm{O}_{5} \mathrm{P}$ requires $M-C_{4} H_{9} O_{2}, 435.2089$ ); $R_{\mathrm{f}} 0.21$ ( $7 \%$ methanol-EtOAc); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3441(\mathrm{OH}), 1438(\mathrm{P}-\mathrm{Ph})$ and $1160(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ 7.7-7.1 $\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 4.68\left(1 \mathrm{H}, \mathrm{d}, J 12.6, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.25(1 \mathrm{H}, \mathrm{d}, J 12.6$, $\left.\mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 3.73\left(1 \mathrm{H}, \mathrm{t}, J 9.2, \mathrm{OCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 3.7-3.5(3 \mathrm{H}, \mathrm{m})$, $2.98(1 \mathrm{H}, \mathrm{t}, J 5.8, \mathrm{CHOH}), 2.6-2.5(3 \mathrm{H}, \mathrm{m}), 2.1-1.85(8 \mathrm{H}, \mathrm{m})$, $0.80(3 \mathrm{H}, \mathrm{t}, J 7.0, \mathrm{Me})$ and $0.50(2 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right)$ $137.3^{-}$(ipso-Ph), 132-126 (m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ph$), 76.1^{+}, 74.0^{+}$, $71.1^{+}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 11.5, \mathrm{CHOH}\right), 69.3^{-}, 63.0^{-}, 33.4^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 70.8\right.$, $\mathrm{PCH}), 31.7^{-}, 30.3^{-}, 29.7^{-}, 26.5^{-}, 25.3^{-}, 22.5^{-}$and $14.0^{+}(\mathrm{Me})$; $\mathrm{m} / \mathrm{z}$ (FAB) $435.2\left(65 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}\right), 202.1$ ( $80, \mathrm{Ph}_{2} \mathrm{POH}$ ) and $91.0(100, \mathrm{Bn})$.

## $\left(2 Z, 4 R^{*}, 5 S^{*}, 6 R^{*}\right)-8$-Benzyloxy-6-diphenylphosphinoyloct-2-ene-1,4,5-triol 48

Cerium trichloride heptahydrate ( $527 \mathrm{mg}, 1.41 \mathrm{mmol}$ ) and enone $47(590 \mathrm{mg}, 1.28 \mathrm{mmol})$ were stirred in ethanol $\left(25 \mathrm{~cm}^{3}\right)$ for 10 min and sodium borohydride ( $336 \mathrm{mg}, 8.84 \mathrm{mmol}$ ) was added. The reaction mixture was stirred for 1 h , quenched with water ( $25 \mathrm{~cm}^{3}$ ), extracted with dichloromethane ( $3 \times 20 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $10 \%$ methanol in EtOAc to give the triol 48 ( $213 \mathrm{mg}, 36 \%$ ) as an oil, $R_{\mathrm{f}} 0.21$ ( $7 \%$ methanol-EtOAc) (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{7} \mathrm{O}_{2}, 379.1474 . \mathrm{C}_{27} \mathrm{H}_{31} \mathrm{O}_{5} \mathrm{P}$ requires $M-\mathrm{C}_{4} \mathrm{H}_{7} \mathrm{O}_{2}$, 379.1463); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3421(\mathrm{OH}), 1654(\mathrm{C}=\mathrm{C})$ and 1438 $(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.80(2 \mathrm{H}$, ddd, $J 1.1,7.7$ and 8.8), $7.6-7.3\left(13 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 5.88(1 \mathrm{H}, \mathrm{td}, J 7.0$ and $\left.11.0, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}\right), 5.48\left(1 \mathrm{H}\right.$, dd, $J 8.3$ and $\left.11.0, \mathrm{CH}_{\mathrm{A}}=\mathrm{CH}_{\mathrm{B}}\right)$, $4.92(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.53\left(1 \mathrm{H}, \mathrm{d}, J 11.8, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.52(1 \mathrm{H}$, $\mathrm{brs}, \mathrm{OH}), 4.32(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.30\left(1 \mathrm{H}, \mathrm{d}, J 11.8, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right)$, $4.13(1 \mathrm{H}, \mathrm{m}), 3.97(1 \mathrm{H}, \mathrm{m}), 3.82(1 \mathrm{H}, \mathrm{t}, J 8.8), 3.55(1 \mathrm{H}, \mathrm{m})$, $3.40(1 \mathrm{H}$, quin, $J 4.0), 2.92(2 \mathrm{H}, \mathrm{m}), 2.35(1 \mathrm{H}, \mathrm{m})$ and 1.85 $(1 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 137.3^{-}$(ipso-Ph), 133-127(m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ph$), 73.1^{-}\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 72.5^{+}\left(\mathrm{d},{ }^{2} J_{\mathrm{PC}} 2.7, \mathrm{CHOH}\right)$, $68.6^{-}\left(\mathrm{d},{ }^{3} J_{\mathrm{PC}} 5.0, \mathrm{CHOBn}\right), 67.0^{+}$(d, $\left.{ }^{3} J_{\mathrm{PC}} 11.6, \mathrm{CHOH}\right)$, $58.3^{-}\left(\mathrm{CH}_{2} \mathrm{OH}\right), 33.8^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 70.0, \mathrm{PCH}\right)$ and $21.0^{-} ; \mathrm{m} / z(\mathrm{FAB})$ $379.1\left(55 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{7} \mathrm{O}_{2}\right), 201.0\left(100, \mathrm{Ph}_{2} \mathrm{PO}\right)$ and 91.1 (95, $\mathrm{Bn})$.

## Catalytic hydrogenation of 48

( $2 Z, 4 R^{*}, 5 R^{*}, 6 S^{*}$ )-8-Benzyloxy-6-diphenylphosphinoyloct-2-ene-1,4,5-triol 48 ( $76 \mathrm{mg}, 0.16 \mathrm{mmol}$ ) and palladium on carbon $(5 \% \mathrm{w} / \mathrm{w}, 8 \mathrm{mg})$ in ethyl acetate $\left(5 \mathrm{~cm}^{3}\right)$ were stirred under an atmosphere of hydrogen ( 1 atm ) for 4 h . The reaction mixture was filtered through Celite and evaporated under reduced pressure to give a crude product. Analysis of the crude product by ${ }^{1} \mathrm{H}$ NMR revealed an $85: 15$ mixture of 49 and starting material.

## (4S,5R,6R,8R)-8-Benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol 58

$m$-Chloroperbenzoic acid ( $202 \mathrm{mg}, 57-85 \%$ by weight, ca. 0.82 mmol ) was added to a stirred solution of the phosphine oxide $57(365 \mathrm{mg}, 0.73 \mathrm{mmol})$ in dry dichloromethane $\left(8 \mathrm{~cm}^{3}\right)$ at $0^{\circ} \mathrm{C}$. The reaction mixture was stirred for 16 h , diluted with dichloromethane ( $10 \mathrm{~cm}^{3}$ ), washed with saturated sodium carbonate solution ( $20 \mathrm{~cm}^{3}$ ), saturated sodium thiosulfate solution ( 20 $\left.\mathrm{cm}^{3}\right)$ and saturated sodium carbonate solution $\left(20 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a residue, which was dissolved in ethanol $\left(10 \mathrm{~cm}^{3}\right)$. Sodium
borohydride ( $173 \mathrm{mg}, 4.6 \mathrm{mmol}$ ) was added to the the reaction mixture which was stirred for 1 h , quenched with water $\left(10 \mathrm{~cm}^{3}\right)$ and the solvent removed under reduced pressure to give a residue which was diluted with water $\left(20 \mathrm{~cm}^{3}\right)$, extracted with dichloromethane ( $3 \times 20 \mathrm{~cm}^{3}$ ), dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with $10 \%$ methanolEtOAc to give the triol $58(180 \mathrm{mg}, 57 \%)$ as an oil, $R_{\mathrm{f}} 0.21(7 \%$ methanol-EtOAc); $[a]_{\mathrm{D}}^{20}-7.3$ (c 0.20 in $\mathrm{CHCl}_{3}$ ) (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}, 435.2135 . \mathrm{C}_{31} \mathrm{H}_{41} \mathrm{O}_{5} \mathrm{P}$ requires $\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}$, 435.2089); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3389(\mathrm{OH})$ and $1423(\mathrm{P}-\mathrm{Ph})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.7-7.3\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 5.8^{*}$ $(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.68\left(1 \mathrm{H}, \mathrm{d}, J 12.1, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.25(1 \mathrm{H}, \mathrm{d}$, $\left.J 12.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.10^{*}(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.6(5 \mathrm{H}, \mathrm{m}), 2.67$ $(1 \mathrm{H}, \mathrm{m}), 2.56(1 \mathrm{H}, \mathrm{m}), 2.0-0.9(11 \mathrm{H}, \mathrm{m})$ and $0.76(3 \mathrm{H}, \mathrm{t}, J 6.8$, $\mathrm{Me}) ; \delta_{\mathrm{C}}\left(50 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 137.4^{-}$(ipso-Ph), 132-128(m, $\mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), $76.3^{+}, 73.5^{+}, 71.2^{+}, 69.8^{-}, 62.7^{-}, 40.2^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 65.8\right.$, PCH), 31.9-, $30.1^{-}, 29.0^{-}, 27.6^{-}, 26.6^{-}, 22.4^{-}$and $14.0^{+}(\mathrm{Me})$; $m / z$ (FAB) $435.2\left(35 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}\right), 216.1$ (90) and 91.0 (100, Bn).

## (4R,5S,6S,8R)-8-Benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol 55

By the same general method, furan ${ }^{1} 54(1.24 \mathrm{~g}, 2.47 \mathrm{mmol})$, $m$-chloroperbenzoic acid ( $780 \mathrm{mg}, 57-85 \%$ by weight, $c a .3 .2$ $\mathrm{mmol})$ and sodium borohydride ( $253 \mathrm{mg}, 6.7 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with $10 \%$ methanol-EtOAc, to give the triol $\mathbf{5 5}(467 \mathrm{mg}$, $36 \%, 40 \%$ based on recovered starting material) as an oil, $R_{\mathrm{f}} 0.21\left(7 \%\right.$ methanol-EtOAc); $[a]_{\mathrm{D}}^{20}+7.1\left(c 0.80\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: $\mathrm{M}^{+}-\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{O}, 465.2195 . \mathrm{C}_{31} \mathrm{H}_{41} \mathrm{O}_{5} \mathrm{P}$ requires $M-$ $\left.C_{3} H_{7} \mathrm{O}, 465.2194\right) ; v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3398(\mathrm{OH}), 1423(\mathrm{P}-\mathrm{Ph})$ and $1222(\mathrm{P}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.9-7.2\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 5.30(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.59\left(1 \mathrm{H}, \mathrm{d}, J 11.1, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right)$, $4.42\left(1 \mathrm{H}, \mathrm{d}, J 11.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.10(1 \mathrm{H}, \mathrm{OH}), 3.7-3.45(5 \mathrm{H}$, $\mathrm{m}), 3.21(1 \mathrm{H}, \mathrm{t}, J 11.9), 2.10(1 \mathrm{H}, \mathrm{m}), 2.0-1.0(11 \mathrm{H}, \mathrm{m})$ and $0.85(3 \mathrm{H}, \mathrm{t}, J 7.3, \mathrm{Me}) ; \delta_{\mathrm{C}}$ ( $100 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) 138.0- (ipso-Ph), 132.5-128 ( $\mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}$ and Ph ), 74.5 ${ }^{+}$, 72.0+ $, 71.4^{-}, 62.9^{-}$(one peak next to O missing), $37.2^{+}$(d, ${ }^{1} J_{\mathrm{PC}}$ 66.2, PCH), $32.6^{-}, 31.2^{-}$, 29.9- $29.4^{-}, 27.2^{-}, 22.8^{-}$and $13.9^{+}(\mathrm{Me}) ; \mathrm{m} / \mathrm{z}$ (FAB) 465.2 ( $45 \%, \mathrm{M}^{+}-\mathrm{C}_{3} \mathrm{H}_{7} \mathrm{O}$ ), 202.1 ( $75, \mathrm{Ph}_{2} \mathrm{POH}$ ), 201.1 ( $70, \mathrm{Ph}_{2} \mathrm{PO}$ ) and $91.1(100, \mathrm{Bn})$. Also obtained was starting material ( $110 \mathrm{mg}, 11 \%$ ), spectroscopically identical to that obtained previously.

## ( $4 R, 5 S, 6 S, 8 S$ )-8-Benzyloxy-8-cyclohexyl-6-diphenylphos-phinoyloctane-1,4,5-triol 61

By the same general method, furan ${ }^{1} \mathbf{6 0}(270 \mathrm{mg}, 0.51 \mathrm{mmol}$, 74:26 mixture of diastereomers), $m$-chloroperbenzoic acid (246 $\mathrm{mg}, 57-85 \%$ by weight, $c a .1 .0 \mathrm{mmol}$ ) and sodium borohydride $(135 \mathrm{mg}, 3.6 \mathrm{mmol})$ gave a crude product which was purified by flash chromatography eluting with $10 \%$ methanol in EtOAc to give the triol $\mathbf{6 1}$ ( $121 \mathrm{mg}, 44 \%, 74: 26$ mixture of diastereomers) as an oil, $R_{\mathrm{f}} 0.30\left(10 \%\right.$ methanol-EtOAc); $[a]_{\mathrm{D}}^{20}+19.0(c 0.05$ in $\mathrm{CHCl}_{3}$ ) (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}, 461.2242 . \mathrm{C}_{33} \mathrm{H}_{43} \mathrm{O}_{5} \mathrm{P}$ requires $\left.M-C_{4} H_{9} O_{2}, 461.2246\right) ; v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3386(\mathrm{OH})$ and $1438(\mathrm{P}-\mathrm{Ph}) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.9-7.2\left(15 \mathrm{H}, \mathrm{m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph$), 5.78\left(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}^{\text {min }}\right), 5.69\left(1 \mathrm{H}, \mathrm{d}, J 3.1, \mathrm{OH}^{\text {maj }}\right), 5.08$ $\left(1 \mathrm{H}, \mathrm{brs}, \mathrm{OH}^{\text {maj }}\right), 4.68\left(1 \mathrm{H}, \mathrm{d}, J 11.1, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}{ }^{\text {min }}\right), 4.63(1 \mathrm{H}$, d, $\left.J 12.1, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}{ }^{\text {maj }}\right), 4.28\left(1 \mathrm{H}, \mathrm{d}, J 12.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}{ }^{\text {maj }}\right), 4.24$ $\left(1 \mathrm{H}, \mathrm{d}, J 11.1, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}{ }^{\text {min }}\right), 3.79(1 \mathrm{H}, \mathrm{m}), 3.28(3 \mathrm{H}, \mathrm{m})$ and 2.1-0.4 ( $14 \mathrm{H}, \mathrm{m}$ ); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 137.5^{-}$(ipso$\left.\mathrm{Ph}^{\text {maj }+ \text { min }}\right), 132-127.5\left(\mathrm{~m}, \mathrm{Ph}_{2} \mathrm{PO}\right.$ and Ph ), $81.7^{+}$(maj), $81.0^{+}$ $(\mathrm{min}), 74.9^{+}(\mathrm{maj}), 73.5^{+}(\mathrm{min}), 72.4^{-}(\mathrm{maj}), 72.1^{+}(\mathrm{maj})$, $71.3^{+}(\mathrm{min}), 62.9^{-}(\min +\mathrm{min}), 40.5^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 66.2, \mathrm{PCH}^{\mathrm{maj}}\right)$, $40.1^{+}\left(\mathrm{d},{ }^{1} J_{\mathrm{PC}} 65.9, \mathrm{PCH}^{\text {min }}\right.$ ) and 31-26 (several peaks); $m / z$ (FAB) $461.2\left(40 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{9} \mathrm{O}_{2}\right), 201.0\left(45, \mathrm{Ph}_{2} \mathrm{PO}\right)$ and 91.1 $(100, \mathrm{Bn})$.

## (4S,5R,6R,8S)-8-Benzyloxy-8-cyclohexyl-6-diphenylphos-phinoyloctane-1,4,5-triol 64

By the same general method, the furan ${ }^{1} \mathbf{6 3}(382 \mathrm{mg}, 0.73 \mathrm{mmol}$, $70: 30$ mixture of isomers), $m$-chloroperbenzoic acid ( 346 mg , $57-85 \%$ by weight, ca. 1.4 mmol ) and sodium borohydride $(194 \mathrm{mg}, 5.1 \mathrm{mmol})$ gave a crude product which was purified by flash chromatography eluting with $10 \%$ methanol in EtOAc to give the triol $\mathbf{6 4}$ ( $186 \mathrm{mg}, 47 \%, 69: 31$ mixture of diastereomers) as an oil, spectroscopically identical to that obtained previously, $[a]_{\mathrm{D}}^{20}-3.6\left(c 1.01\right.$ in $\left.\mathrm{CHCl}_{3}\right)$.

## ( $E$ )-8-Benzyloxyoct-5-ene-1,4-diol 50

Potassium hydroxide ( $79 \mathrm{mg}, 2.1 \mathrm{mmol}$ ) and ( $4 R^{*}, 5 S^{*}, 6 R^{*}$ )-8-benzyloxy-6-diphenylphosphinoyloctane-1,4,5-triol 49 (201 $\mathrm{mg}, 0.43 \mathrm{mmol})$ were stirred in DMSO $\left(10 \mathrm{~cm}^{3}\right)$ at $55^{\circ} \mathrm{C}$ for 2 h . The reaction mixture was quenched with water $\left(10 \mathrm{~cm}^{3}\right)$, extracted with $\mathrm{Et}_{2} \mathrm{O}\left(3 \times 10 \mathrm{~cm}^{3}\right)$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give a crude product which was purified by flash chromatography eluting with EtOAc to give the alkene $50(37 \mathrm{mg}, 35 \%)$ as an oil, $R_{\mathrm{f}} 0.77(10 \%$ methanol-EtOAc) (Found: $\mathrm{MH}^{+}, 251.1626 . \mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{3}$ requires MH, 251.1647); $v_{\max } / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3450(\mathrm{OH})$ and $1654(\mathrm{C}=\mathrm{C})$; $\delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.4-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.66(1 \mathrm{H}, \mathrm{td}$, $J 6.4$ and $15.8, \mathrm{CH}=\mathrm{CHCH} 2), 5.55(1 \mathrm{H}, \mathrm{dd}, J 6.6$ and 15.5 , $\left.\mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.51\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.09(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOH}), 3.62$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{O}\right), 3.49\left(2 \mathrm{H}, \mathrm{t}, J 6.2, \mathrm{CH}_{2} \mathrm{O}\right), 2.61(2 \mathrm{H}, \mathrm{br} \mathrm{s}$, $2 \times \mathrm{OH}), 2.38\left(2 \mathrm{H}, \mathrm{q}, J 6.6, \mathrm{CH}=\mathrm{CHCH}_{2}\right)$ and $1.62(4 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 138.4^{+}$(ipso- Ph$), 134.9^{+}(\mathrm{CH}=\mathrm{CH})$, $128.8^{+}, 127.9^{+}, 72.9^{-}, 72.6^{+}(\mathrm{CHOH}), 69.7^{-}, 62.8^{-}, 34.2^{-}, 32.6^{-}$ and $28.4^{-} ; m / z(\mathrm{FAB}) 251.2\left(20 \%, \mathrm{MH}^{+}\right)$and $91.1(100, \mathrm{Bn})$.

## (4S,5E,8R)-8-Benzyloxydodec-5-ene-1,4-diol 59

By the same general method, ( $4 S, 5 R, 6 R, 8 R$ )-8-benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol $\mathbf{5 8}(129 \mathrm{mg}, \quad 0.25$ mmol ) and potassium hydroxide ( $45 \mathrm{mg}, 1.18 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with EtOAc to give the alkene $59(20.6 \mathrm{mg}, 27 \%)$ as an oil, $R_{\mathrm{f}} 0.38$ ( EtOAc ); $[a]_{\mathrm{D}}^{20}+4.7$ (c 0.91 in $\mathrm{CDCl}_{3}$ ) (Found: $\mathrm{M}^{+}-\mathrm{OH}, 289.2173 . \mathrm{C}_{19} \mathrm{H}_{28} \mathrm{O}_{3}$ requires $\mathrm{M}-\mathrm{OH}, 289.2167$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3411(\mathrm{br} \mathrm{s}, \mathrm{OH}), 1642,1602$ and $1560(\mathrm{C}=\mathrm{C}$ and Ph$) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.38-7.23(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.68$ ( $1 \mathrm{H}, \mathrm{td}, J 7.0$ and $15.5, \mathrm{CH}=\mathrm{CHCH}_{2}$ ), $5.55(1 \mathrm{H}, \mathrm{dd}, J 6.8$ and $\left.15.5, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.52\left(1 \mathrm{H}, \mathrm{d}, J 11.6, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.48(1 \mathrm{H}$, d, $\left.J 11.6, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.10(1 \mathrm{H}, \mathrm{q}, J 6.4, \mathrm{CHOH}), 3.63(2 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}_{2} \mathrm{OH}$ ), 3.42 ( 1 H , quin, $J 7.1, \mathrm{CHOBn}$ ), $2.30(2 \mathrm{H}, \mathrm{t}, J 6.2$, $\left.\mathrm{CH}=\mathrm{CHCH}_{2}\right), 1.73(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 1.7-1.2(10 \mathrm{H}, \mathrm{m})$ and 0.90 ( $3 \mathrm{H}, \mathrm{t}, J 6.7, \mathrm{Me}$ ); $\delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right.$ ) $138.9^{-}$(ipso- Ph ), $135.3^{+}(\mathrm{CH}=\mathrm{CH}), 128.3^{+}, 127.9^{+}, 127.7^{+}, 127.5^{+}, 78.6^{+}, 72.7^{+}$, $70.9^{-}, 62.9^{-}, 36.5^{-}, 34.2^{-}, 33.5^{-}, 28.9^{-}, 27.6^{-}, 22.7^{-}$and $14.1^{+}$ (Me); $m / z$ (FAB) $289.2\left(20 \%, \mathrm{M}^{+}-\mathrm{OH}\right)$ and 91.1 (100, Bn). Integration of the 500 MHz NMR spectrum of the Mosher's diester ${ }^{31}$ of this material indicated a ratio of $(4 S):(4 R)$ stereoisomers of $89: 11$.

## (4R,5E,8R)-8-Benzyloxydodec-5-ene-1,4-diol 56

By the same general method, $(4 R, 5 S, 6 S, 8 R)-8$-benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol $\mathbf{5 5}(174 \mathrm{mg}, 0.34$ mmol ) and potassium hydroxide ( $61 \mathrm{mg}, 1.61 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with EtOAc to give the alkene $56(41.2 \mathrm{mg}, 41 \%)$ as an oil, $R_{\mathrm{f}} 0.38(\mathrm{EtOAc}) ;[a]_{\mathrm{D}}^{20}+10.1$ (c 0.91 in $\mathrm{CDCl}_{3}$ ) (Found: $\mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}_{2}$, 216.1515. $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{O}_{3}$ requires $M-\mathrm{C}_{4} H_{10} O_{2}$, 216.1514); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3422$ (br s, OH ), 1642, 1602 and $1560(\mathrm{C}=\mathrm{C}$ and Ph$) ; \delta_{\mathrm{H}}\left(400 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 7.36-7.23(5 \mathrm{H}, \mathrm{m}$, $\mathrm{Ph}), 5.66\left(1 \mathrm{H}, \mathrm{td}, J 7.0\right.$ and $\left.15.4, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 5.53(1 \mathrm{H}, \mathrm{dd}$, $J 6.7$ and $\left.15.4, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.52\left(1 \mathrm{H}, \mathrm{d}, J 11.7, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right)$, $4.46\left(1 \mathrm{H}, \mathrm{d}, J 11.7, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right), 4.08(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOH}), 3.62$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.44(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOBn}), 2.55(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$,
$2.29\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 1.65-1.2(10 \mathrm{H}, \mathrm{m})$ and $0.88(3 \mathrm{H}$, $\mathrm{t}, J 6.7, \mathrm{Me}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 138.8^{-}$(ipso-Ph), $135.3^{+}$ $(\mathrm{CH}=\mathrm{CH}), 128.3^{+}, 127.8^{+}, 127.7^{+}, 127.5^{+}, 78.6^{+}, 72.7^{+}, 70.9^{-}$, $62.8^{-}, 36.5^{-}, 34.3^{-}, 33.5^{-}, 28.9^{-}, 27.6^{-}, 22.8^{-}$and $14.1^{+}(\mathrm{Me}) ;$ $m / z 216.2\left(45 \%, \mathrm{M}^{+}-\mathrm{C}_{4} \mathrm{H}_{10} \mathrm{O}_{2}\right), 105.0(80)$ and 91.1 (100, Bn$)$. Integration of the 500 MHz NMR spectrum of the Mosher's diester ${ }^{31}$ of this material indicated a ratio of $(4 R):(4 S)$ stereoisomers of $74: 26$.

## (4R,5E,8S)-8-Benzyloxy-8-cyclohexyloct-5-ene-1,4-diol 62

By the same general method, ( $4 R, 5 S, 6 S, 8 S$ )-8-benzyloxy-8-cyclohexyl-6-diphenylphosphinoyloctane-1,4,5-triol 61 (105 $\mathrm{mg}, 0.19 \mathrm{mmol}$ ) and potassium hydroxide ( $34 \mathrm{mg}, 0.89 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with EtOAc, to give the alkene $62(21.2 \mathrm{mg}$, $34 \%)$ as an oil, $R_{\mathrm{f}} 0.36(\mathrm{EtOAc}) ;[a]_{\mathrm{D}}^{20}+6.4\left(c 0.89\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: $\mathrm{MNa}^{+}, 355.2261 . \mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M N a, 355.2249$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3407(\mathrm{OH})$ and $1662(\mathrm{C}=\mathrm{C}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 7.4-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.76(1 \mathrm{H}, \mathrm{td}, J 7.1$ and 15.4 , $\left.\mathrm{CH}=\mathrm{CHCH}_{2}\right), 5.51\left(1 \mathrm{H}, \mathrm{dd}, J 6.7\right.$ and $\left.15.4, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.52$ $\left(1 \mathrm{H}, \mathrm{d}, J 11.5, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.40\left(1 \mathrm{H}, \mathrm{d}, J 11.5, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right)$, $4.11(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOH}), 3.62\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.19(1 \mathrm{H}, \mathrm{q}, J$ 5.3, CHOBn ), $2.35\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 1.9-0.9(15 \mathrm{H}, \mathrm{m}) ; \delta_{\mathrm{C}}$ ( $100 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) $138.9^{-}$(ipso- Ph ), $134.9^{+}(\mathrm{CH}=\mathrm{CH}), 128.2^{+}$, $127.9^{+}, 127.7^{+}, 127.5^{+}, 83.3^{+}, 72.7^{+}, 71.8^{-}, 62.9^{-}, 41.0^{+}$ (CHCHOBn), 34.2-, 33.5-, 33.4 ${ }^{-}$, 29.0 $0^{-}$, 28.9 $9^{-}$, $28.7^{-}, 26.6^{-}$ and $26.3^{-} ; ~ m / z 203.1$ ( $100 \%$, CHOBn $^{\text {c }} \mathrm{Hex}$ ), 111.1 (90) and 91.1 $(100, B n)$.

## (4S,5E,8S)-8-Benzyloxy-8-cyclohexyloct-5-ene-1,4-diol 65

By the same general method, $(4 S, 5 R, 6 R, 8 S)$-8-benzyloxy-8-cyclohexyl-6-diphenylphosphinoyloctane-1,4,5-triol 64 (140 $\mathrm{mg}, 0.25 \mathrm{mmol}$ ) and potassium hydroxide ( $44 \mathrm{mg}, 1.16 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with EtOAc to give the alkene $65(20.6 \mathrm{mg}$, $25 \%$ ) as an oil, $R_{\mathrm{f}} 0.36(\mathrm{EtOAc}) ;[a]_{\mathrm{D}}^{20}+8.6\left(c 0.45\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: $\mathrm{MNa}^{+}, 355.2261 . \mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{3}$ requires $M N a, 355.2249$ ); $v_{\text {max }} / \mathrm{cm}^{-1}\left(\mathrm{CHCl}_{3}\right) 3413(\mathrm{OH})$ and $1669(\mathrm{C}=\mathrm{C}) ; \delta_{\mathrm{H}}(400 \mathrm{MHz}$; $\left.\mathrm{CDCl}_{3}\right) 7.4-7.25(5 \mathrm{H}, \mathrm{m}, \mathrm{Ph}), 5.76(1 \mathrm{H}, \mathrm{td}, J 7.1$ and 15.4 , $\left.\mathrm{CH}=\mathrm{CHCH}_{2}\right), 5.51\left(1 \mathrm{H}, \mathrm{dd}, J 6.7\right.$ and $\left.15.4, \mathrm{CH}=\mathrm{CHCH}_{2}\right), 4.52$ $\left(1 \mathrm{H}, \mathrm{d}, J 11.5, \mathrm{PhCH}_{\mathrm{A}} \mathrm{H}_{\mathrm{B}}\right), 4.49\left(1 \mathrm{H}, \mathrm{d}, J 11.5, \mathrm{PhCH}_{\mathrm{A}} H_{\mathrm{B}}\right)$, $4.11(1 \mathrm{H}, \mathrm{m}, \mathrm{CHOH}), 3.62\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.19(1 \mathrm{H}, \mathrm{q}$, $J 5.3, \mathrm{CHOBn}), 2.35\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}=\mathrm{CHCH}_{2}\right)$ and 1.9-0.9 $(15 \mathrm{H}$, $\mathrm{m}) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz} ; \mathrm{CDCl}_{3}\right) 138.9^{-}$(ipso-Ph), $135.0^{+}(\mathrm{CH}=\mathrm{CH})$, $128.2^{+}, 127.9^{+}, 127.8^{+}, 127.5^{+}, 83.4^{+}, 72.7^{+}, 71.8^{-}, 62.9^{-}, 41.0^{+}$
 $26.4^{-}$and $26.3^{-}$; $m / z 203.1$ ( $100 \%, \mathrm{CHOBn}^{\mathrm{c}} \mathrm{Hex}$ ), 111.1 (90) and $91.1(100, \mathrm{Bn})$.

## (4R,5E,8R)-8-Benzyloxydodec-5-ene-1,4-diol 56

By the same general method, $(4 R, 5 S, 6 R, 8 R)$-8-benzyloxy-6-diphenylphosphinoyldodecane-1,4,5-triol 68 ( $306 \mathrm{mg}, 0.58$ mmol ) and potassium hydroxide ( $107 \mathrm{mg}, 2.81 \mathrm{mmol}$ ) gave a crude product which was purified by flash chromatography eluting with EtOAc to give the alkene $\mathbf{5 6}(41.3 \mathrm{mg}, 23 \%)$ as an oil, spectroscopically identical to that obtained previously.

## References

1 A. Nelson and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1999, preceding paper
2 For example, see: (a) H. J. Mitchell and S. Warren, Tetrahedron Lett., 1996, 37, 2105; (b) C. Guéguen, P. O'Brien, S. Warren and P. Wyatt, J. Organomet. Chem., 1997, 529, 279; (c) C. Guéguen, P. O'Brien, H. R. Powell, P. R. Raithby and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1998, 3405; (d) J. Clayden, E. W. Collington, R. B. Lamont and S. Warren, Tetrahedron Lett., 1993, 34, 2203; (e) J. Clayden and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1998, 2923

3 (a) D. Hall, A.-F. Sévin and S. Warren, Tetrahedron Lett., 1991, 32, 7123 (the assignment of syn stereochemistry to the epoxides in this paper is incorrect; the epoxide oxygen should be anti to the alcohol
as implied here); (b) J. P. Clayden, PhD Thesis, University of Cambridge, 1992; (c) H. J. Mitchell, PhD Thesis, University of Cambridge, 1995.
4 H. Mitchell, A. Nelson and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1999, 1899.
5 J. Clayden and S. Warren, Angew. Chem., Int. Ed. Engl., 1996, 35, 241.

6 A.-F. Sévin, thèse, Docteur en Sciences, Université de Paris Sud (Orsay), 1991.
7 R. S. Torr and S. Warren, J. Chem Soc., Perkin Trans. 1, 1983, 1169.
8 J. Clayden, A. B. McElroy and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1995, 1913.
9 (a) G. V. Bindu Madhaven, D. P. C. McGee, R. M. Rydzewski, R. Boehme, J. C. Martin and E. J. Prisbe, J. Med. Chem., 1998, 31, 1798; (b) K. Biggadike, A. D. Borthwick, A. M. Exall, B. E. Kirk, S. M. Roberts and P. Youds, J. Chem. Soc., Chem. Commun., 1987, 1083; (c) T. Seita, M. Kinoshita and M. Imoto, Bull. Chem. Soc. Jpn., 1973, 46, 1572; (d) H. Baumgartner, C. Marschner, P. Pucher and H. Griengl, Tetrahedron Lett., 1991, 32, 611
10 P. H. J. Carlsen, T. Katsuki, V. S. Martin and K. B. Sharpless, J. Org. Chem., 1981, 46, 3936.

11 (a) D. L. Hughes, Org. React., 1992, 42, 335; (b) J. Mulzer and C. Brand, Tetrahedron, 1986, 42, 5961; (c) A. V. Rama Rao, D. Subhas Bose, M. K. Gurjar and T. Ravindranathan, Tetrahedron, 1989, 45, 7031; (d) G. L. Grunewald, V. M. Paradkar, D. Pazhenchevsky, M. A. Pleiss, D. L. Sall, W. L. Seibel and M. J. Reitz, J. Org. Chem., 1983, 48, 2321; (e) H. Ida, N. Yamazaki and C. Kibayashi, J. Chem. Soc., Chem. Commun., 1987, 746.

12 P. F. Schuda, M. B. Cichowicz and M. R. Heimann, Tetrahedron Lett., 1983, 24, 3829
13 (a) A. Nelson, P. O'Brien and S. Warren, Tetrahedron Lett., 1995, 36, 2685; (b) A. Nelson and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1997, 2645.
14 (a) P. Theisen and C. H. Heathcock, J. Org. Chem., 1988, 53, 2374; (b) I. Paterson, R. D. Norcross, R. A. Ward, P. Romea and M. A. Lister, J. Am. Chem. Soc., 1994, 116, 11287.

15 J. Clayden and S. Warren, J. Chem. Soc., Perkin Trans. 1, 1993, 2913.

16 B. H. Lipshutz, Chem. Rev., 1986, 86, 795.
17 (a) C. Poirel, P.-Y. Renard and J.-Y. Lallemand, Tetrahedron Lett., 1994, 35, 6485; (b) S. F. Martin and P. W. Zinke, J. Org. Chem., 1991, 56, 6600; (c) F. E. Ziegler and R. T. Wester, Tetrahedron Lett., 1984, 25, 617; (d) P. DeShong, D. M. Simpson and M.-T. Lin, Tetrahedron Lett., 1989, 30, 2885; (e) J. Raczko, A. Golebiowski, J. W. Krajewski, P. Gluzinski and J. Jurczak, Tetrahedron Lett., 1990, 31, 3797; ( f ) P. M. Carbateas and G. L. Williams, J. Heterocycl. Chem., 1974, 11, 819; (g) R. Antonioletti, L. Arista, F. Bonadies, L. Locati and A. Scettri, Tetrahedron Lett., 1993, 34, 7089; (h) J. M. J. Williams, Synlett, 1996, 705; (i) S. F. Martin and P. W. Zinke, J. Am. Chem. Soc., 1989, 111, 2311; (j) P. DeShong, R. E. Waltermire and H. L. Ammon, J. Am. Chem. Soc., 1988, 110, 1901; (k) R. C. D. Brown and P. J. Kocienski, Synlett, 1994, 417; (l) S. F. Martin, W.-C. Lee, G. J. Pacofsky, R. P. Gist and R. A. Mulhern, J. Am. Chem. Soc., 1994, 116, 4674.
18 T. Kametani, M. Tsubuki and T. Honda, Chem. Pharm. Bull., 1988, 36, 3706.
19 (a) T. Honda, the late T. Kametani, K. Kanai, Y. Tatsuzaki and M. Tsubuki, J. Chem. Soc., Perkin Trans. 1, 1990, 1733; (b) P. G. Sammes and D. Thetford, J. Chem. Soc., Perkin Trans. 1, 1988, 111; (c) M. Kasakabe, T. Kitano, Y. Kobayashi and F. Sato, J. Org. Chem., 1989, 54, 2085.
20 T. Nakata, T. Tanaka and T. Oishi, Tetrahedron Lett., 1983, 24, 2653.

21 (a) M. Chérest, H. Felkin and N. Prudent, Tetrahedron Lett., 1968, 2199; (b) N. T. Anh and O. Eisenstein, Nouv. J. Chem., 1977, 1, 61; (c) N. T. Anh, Top. Curr. Chem., 1980, 88, 145; (d) M. N. Paddon-Row, N. G. Rondan and K. N. Houk, J. Am. Chem. Soc., 1983, 104, 7162.
22 (a) J.-L. Luche, J. Am. Chem. Soc., 1978, 100, 2226; (b) A. L. Gemal and J.-L. Luche, J. Am. Chem. Soc., 1981, 103, 5454.
23 (a) G. Hutton, T. Jolliff, H. J. Mitchell and S. Warren, Tetrahedron Lett., 1995, 36, 7905; (b) H. Fujii, K. Oshima and K. Utimoto, Chem. Lett., 1992, 967; (c) S. Matsubara, H. Takahashi and K. Utimoto, Chem. Lett., 1992, 2173; (d) K. Li, L. G. Hamann and M. Koreeda, Tetrahedron Lett., 1992, 33, 6569; (e) S. E. Denmark and J. Amburgey, J. Am. Chem. Soc., 1993, 115, 10386; (f) C. Agami, F. Couty and C. Lequesne, Tetrahedron, 1995, 51, 4043.
24 (a) E. W. Collington, J. G. Knight, C. J. Wallis and S. Warren, Tetrahedron Lett., 1989, 30, 877; (b) K. M. Brown, N. J. Lawrence, J. Liddle, F. Muhammad and D. A. Jackson, Tetrahedron Lett., 1994, 35, 6733; (c) B. Lythgoe, T. A. Moran, M. E. N. Nambudiry and S. Ruston, J. Chem. Soc., Perkin Trans. 1, 1976, 2386;
(d) E. Vedejs, J. B. Campbell, R. C. Gadwood, J. D. Rodgers, K. L. Spear and Y. Watanabe, J. Org. Chem., 1982, 47, 1534; (e) J. M. Clough and G. Pattenden, Tetrahedron Lett., 1978, 4159; (f) R. E. Dolle and K. C. Nicolaou, J. Chem. Soc., Chem. Commun., 1985, 1016; (g) W. G. Dauben, R. R. Ollmann, A. S. Funhoff, S. S. Leung, A. W. Norman and J. E. Bishop, Tetrahedron Lett., 1991, 32, 4643; (h) W. G. Dauben and L. J. Greenfield, J. Org. Chem., 1992, 57, 1597; (i) K. L. Perlman and H. F. de Luca, Tetrahedron Lett., 1992, 33, 2937; (j) B. Lythgoe, Chem. Rev., 1980, 80, 449
25 Preliminary communication: A. Nelson and S. Warren, Tetrahedron Lett., 1998, 39, 1633.
26 (a) E. G. Neeland, A. Sharadenda and L. Weiler, Tetrahedron Lett., 1996, 37, 5069; (b) S. J. Stanway and E. J. Thomas, J. Chem. Soc., Chem. Commun., 1994, 285
27 (a) I. Ohtani, T. Kusumi, Y. Kashman and H. Kakisawa, J. Am Chem. Soc., 1991, 113, 4092; (b) J. A. Dale and H. S. Mosher, J. Am. Chem. Soc., 1973, 95, 512; (c) D. Parker, Chem. Rev., 1991, 91, 1441; (d) S. Yamaguch, in Asymmetric Synthesis, vol. 1, ch. 7, ed. J. D. Morrison, Academic Press, New York, 1983.

28 T. Pehk, E. Lippma, M. Lopp, A. Paju, B. C. Borer and R. J. K. Taylor, Tetrahedron: Asymmetry, 1993, 4, 1527.
29 (a) E. J. Thomas, Chem. Commun., 1997, 411; (b) J. S. Carey, T. S. Coulter, D. J. Hallett, R. J. Maguire, A. H. McNeill, S. J. Stanway, A. Teerwutgulrag and E. J. Thomas, Pure App. Chem., 1996, 68, 707; (c) E. J. Thomas and A. H. McNeill, Synthesis, 1994, 322; (d) E. J. Thomas and D. J. Hallett, J. Chem. Soc., Chem. Commun., 1995, 657; (e) E. J. Thomas and S. J. Stanway, J. Chem. Soc., Chem. Commun., 1994, 285; ( $f$ ) E. J. Thomas and J. S. Carey, Synlett, 1992, 585.
30 A. I. Vogel, Textbook of Practical Chemistry, 4th edn., Longman, London, 1978, p. 291.
31 J. A. Dale, D. L. Dull and H. S. Mosher, J. Org. Chem., 1969, 34, 2543.


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[^1]:    $\ddagger$ In the light of previous work, these low-yielding eliminations were, perhaps, to be expected (ref. 8).

[^2]:    ** The reduction of similar methyl acetals was completely 1,2-regioselective (ref. 19).

[^3]:    $\ddagger$ Mosher's correlation has been applied to similar secondary allylic alcohols (ref. 28).

